

# High Specific Energy Batteries for Future Missions to Icy Worlds

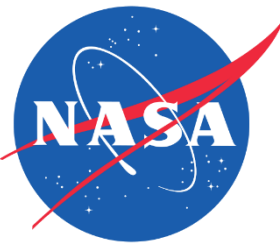
John-Paul Jones

Electrochemical Technologies Group (3463)

April 6<sup>th</sup>, 2017

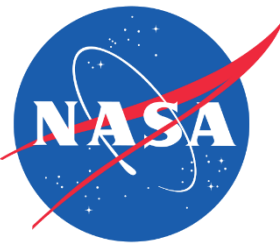
Jet Propulsion Laboratory, California Institute of Technology

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Government sponsorship acknowledged



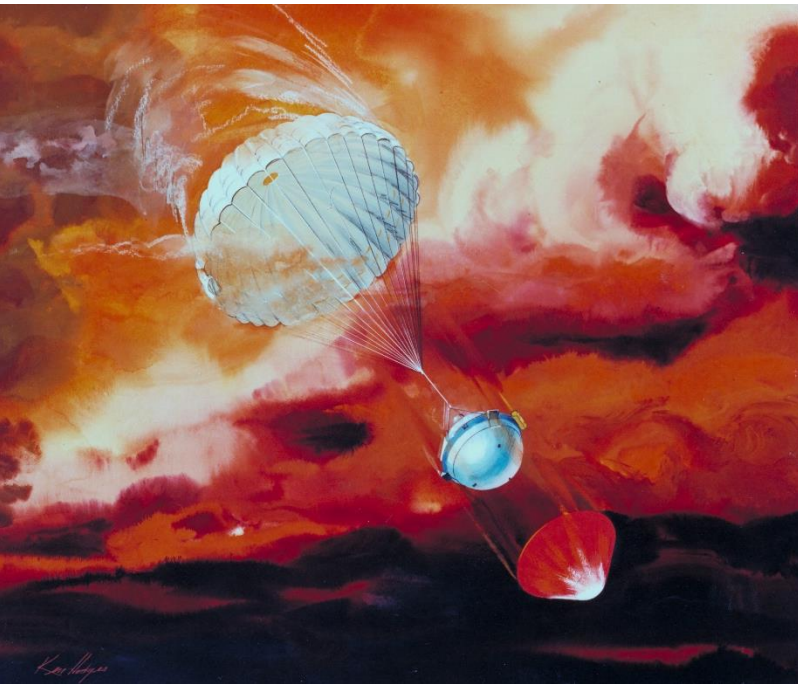
# Outline

- Lithium primary battery performance testing
  - Rate testing
  - Temperature testing
  - Radiation testing
  - Storage testing
- Experimental lithium primary cell testing
  - Dry heat microbial reduction
- Experimental lithium-ion cell testing
  - Plating during charging at low temperature

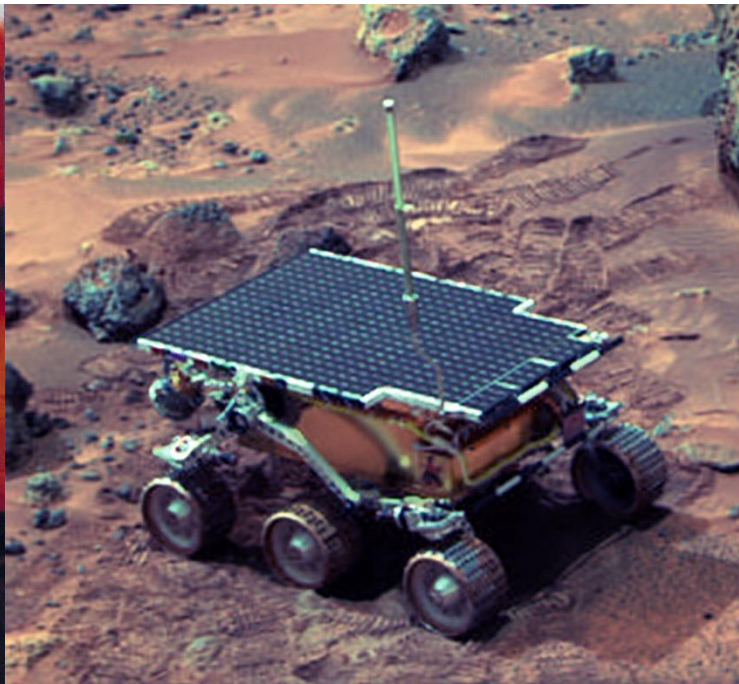


# Three Examples of Primary Batteries for Space

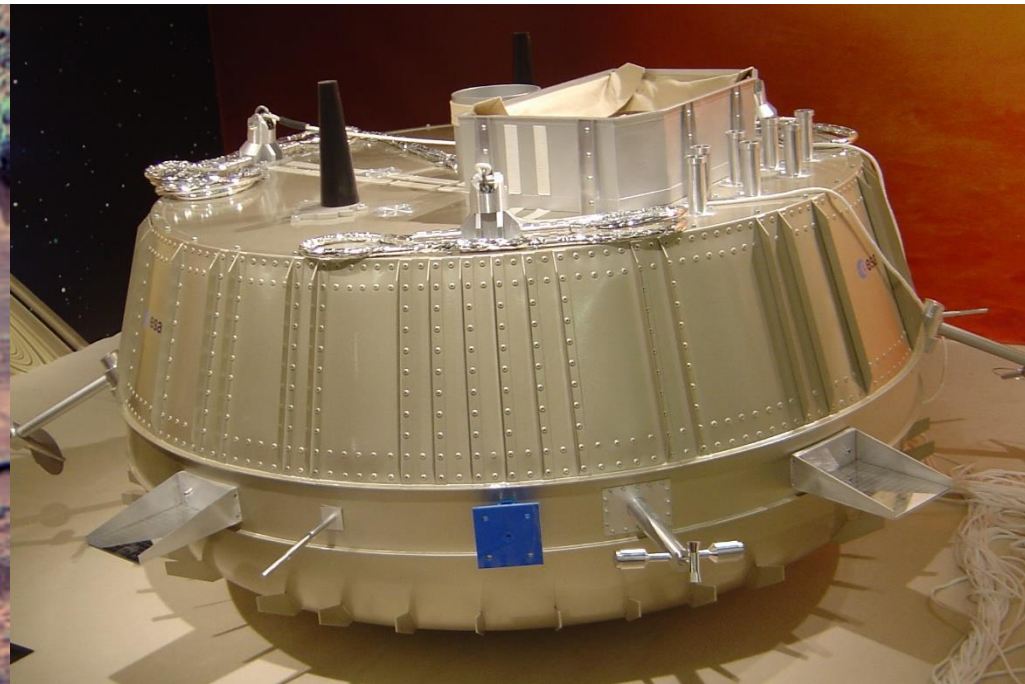
Galileo Probe 1989:  $\text{Li}/\text{SO}_2$   
~580 Wh  
58 minutes



Sojourner Rover 1996:  $\text{Li}/\text{SOCl}_2$   
432 Wh  
56 days (PV + battery)



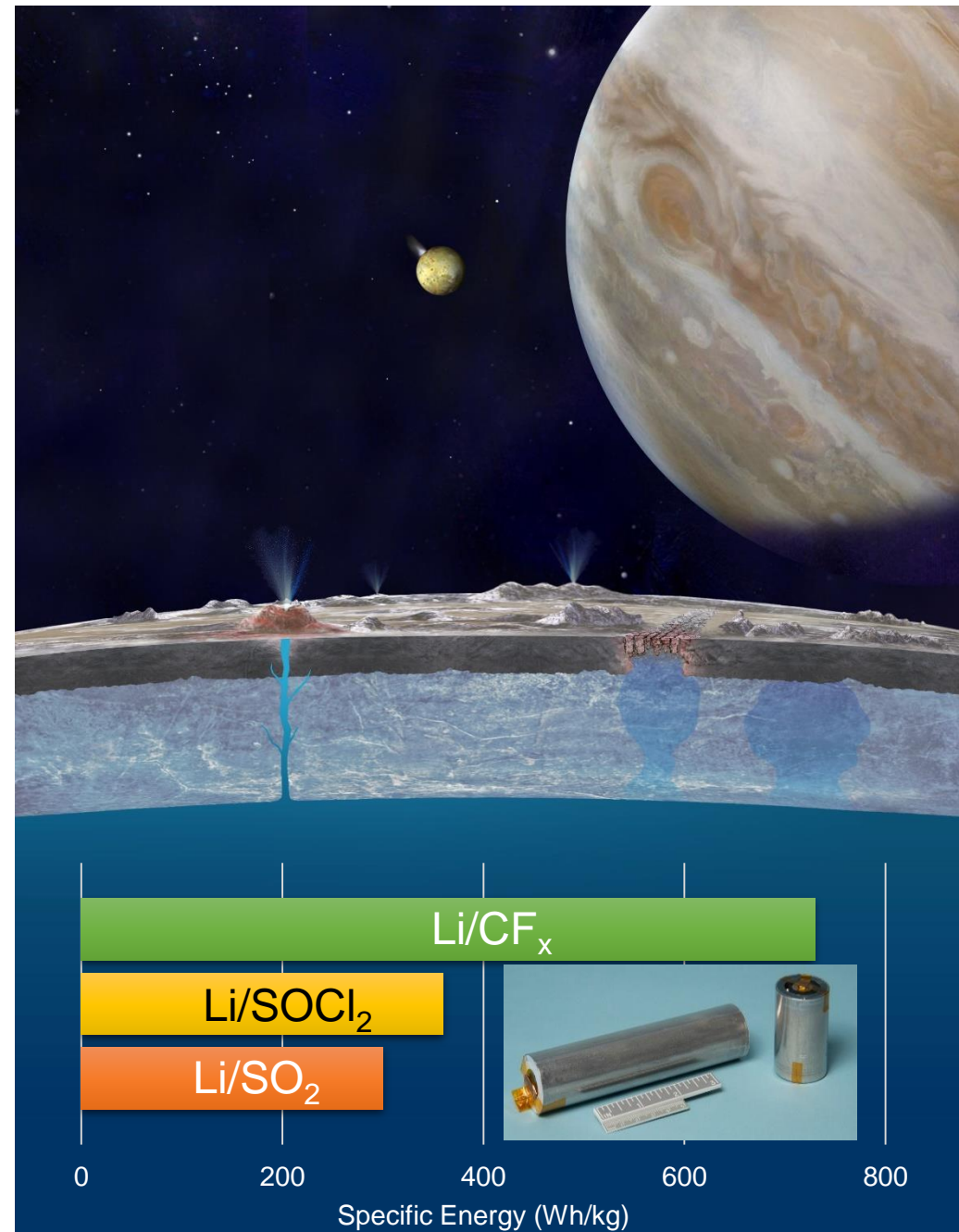
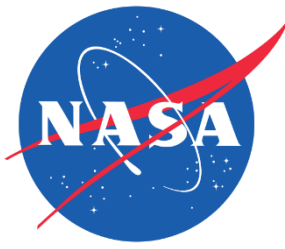
Huygens Probe 2004:  $\text{Li}/\text{SO}_2$   
~2700 Wh  
153 minutes



**A Europa lander could require at least 480 hours of operation on battery power alone**

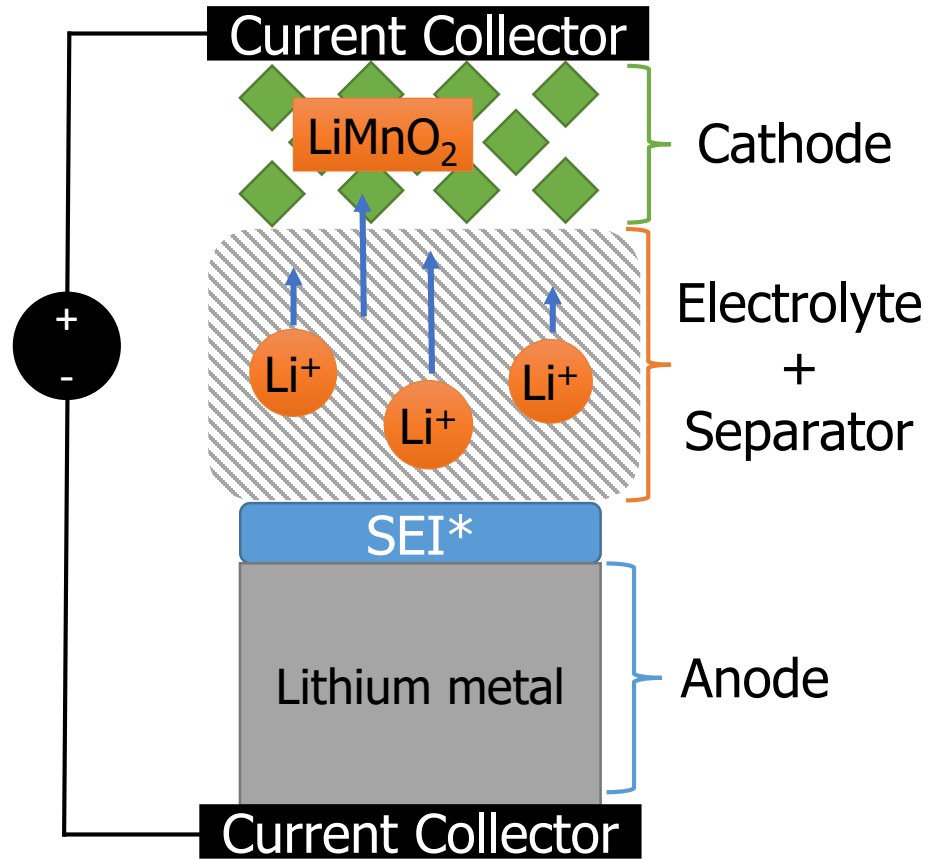
# Future Mission Needs

- Jupiter and Saturn's moons are attractive targets for future surface missions
- Extreme environments present significant challenges:
  - Europa:  $-171\text{ }^{\circ}\text{C}$
  - Enceladus:  $-198\text{ }^{\circ}\text{C}$
  - Titan:  $-180\text{ }^{\circ}\text{C}$
- Planetary protection requirements stringent for landers
- Specific energy must be high
- **Long duration missions in harsh environments demand new battery technology**





# Battery basics



Lithium cell during discharge

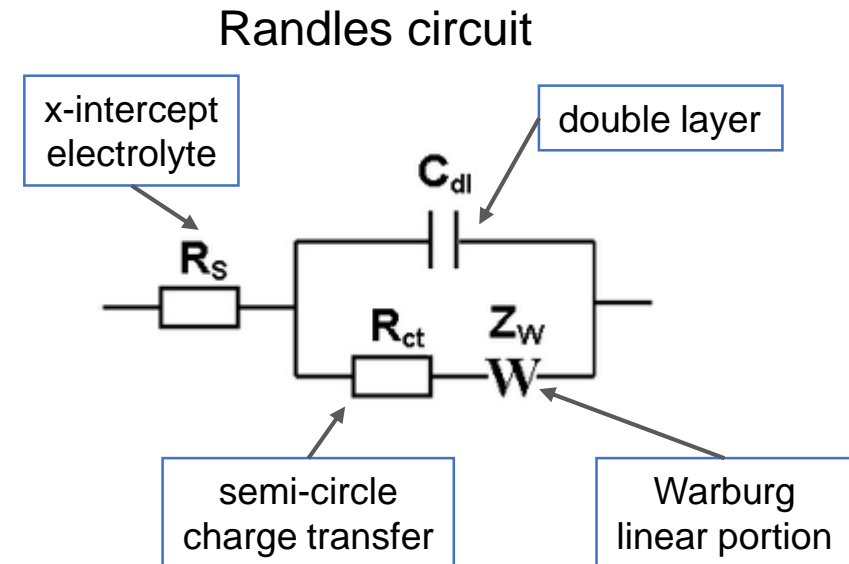
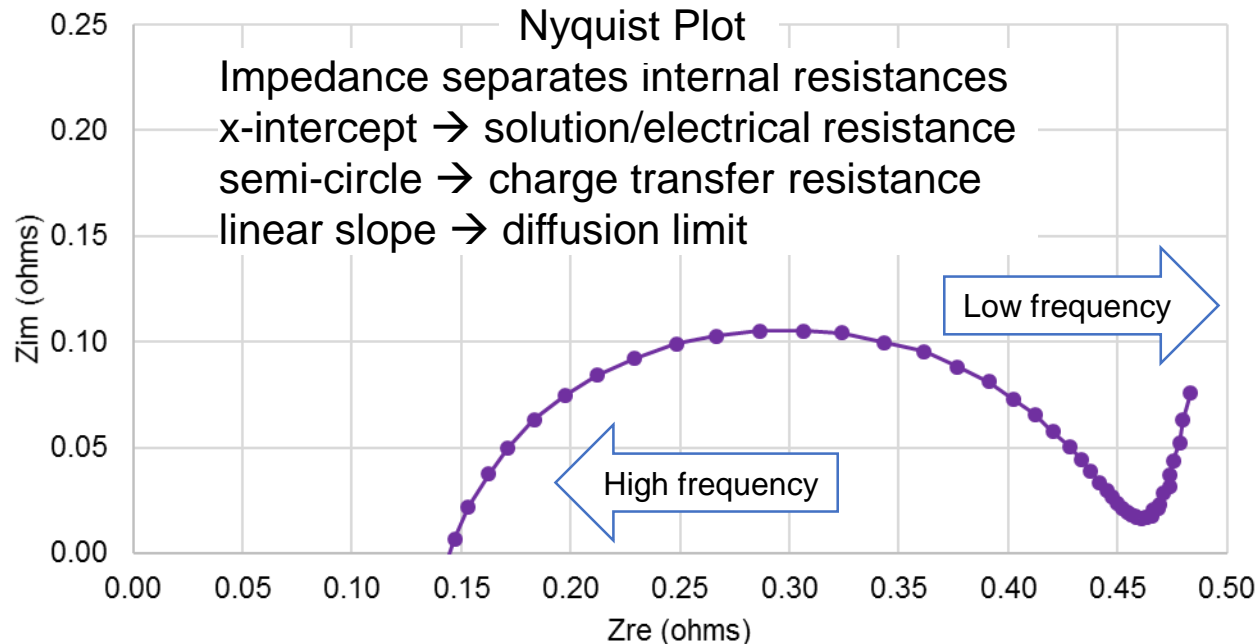
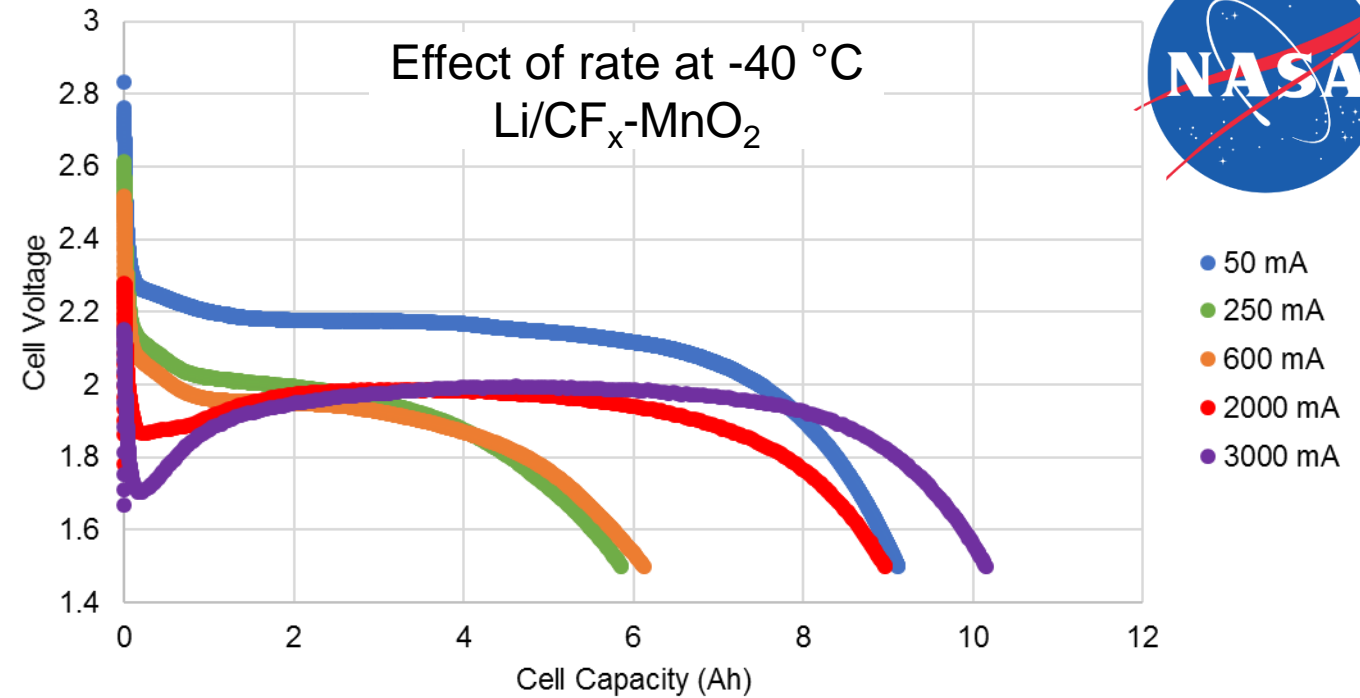
Metrics for comparison:

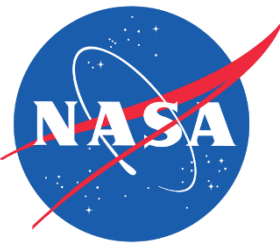
- Capacity (amp-hours)
- Voltage
- Energy (watt-hours)
- **Specific Energy (watt-hours/kg)**
- Volumetric energy density (watt-hours/L)
- Rate capability (c-rate)
- Low/high temperature performance
- Self discharge

\* SEI = Solid Electrolyte Interphase, which protects the electrolyte from the highly reactive electrodes

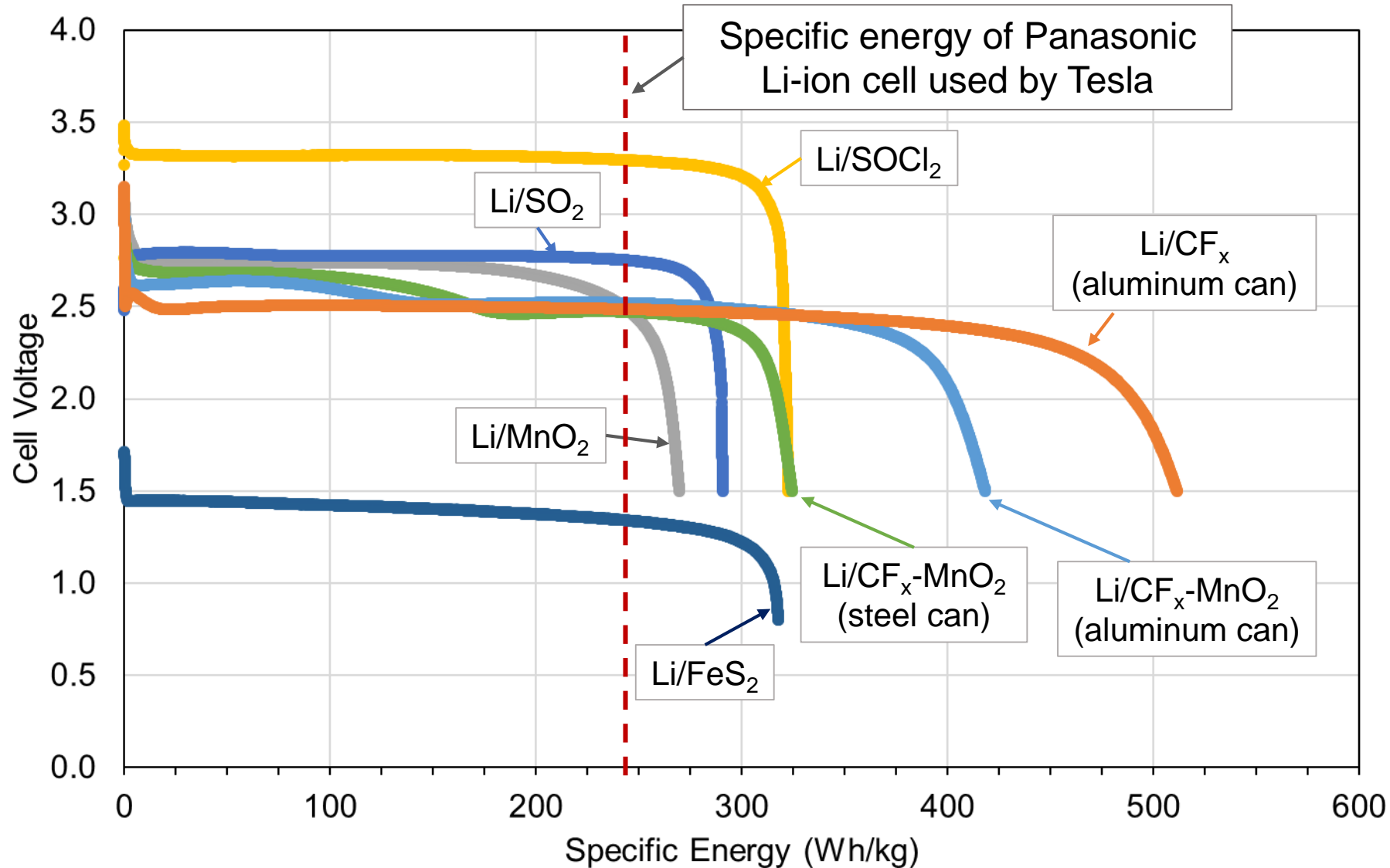
# Testing methods

- Discharge/cycling performance
  - Rate (50, 250 and 600 mA)
  - Temperature (-40 to +21 °C)
- Impedance analysis
  - Apply range of frequencies to cell
  - “static” conditions

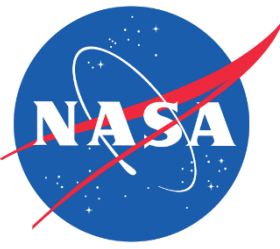




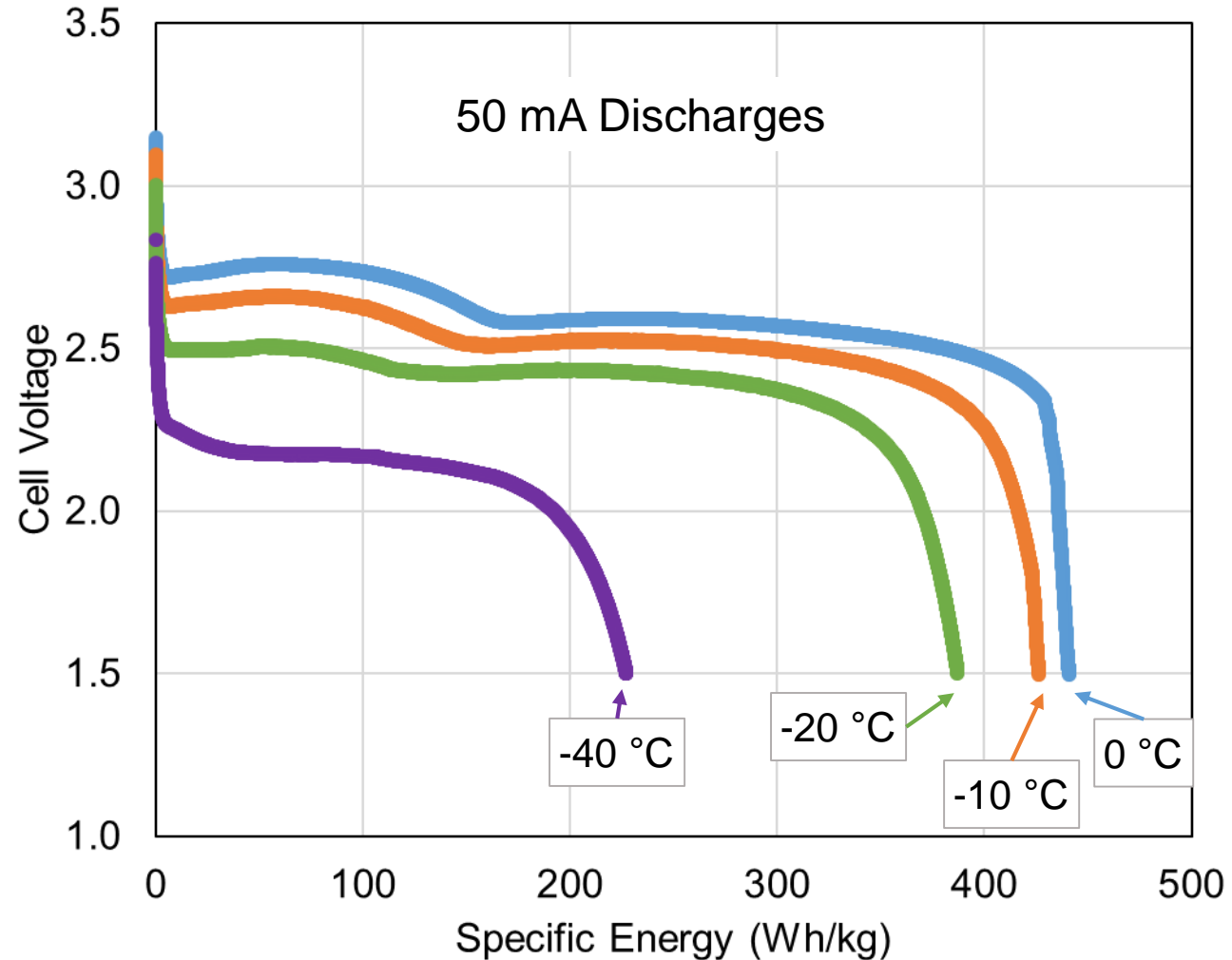
# D-size Lithium Primary Battery Comparison



- Discharged at the same condition
- 0 °C, 250 mA
- Li/FeS<sub>2</sub> discharged at 100 mA due to size (AA instead of D)
- 2 Li/FeS<sub>2</sub> cells could be connected in series to provide comparable voltage

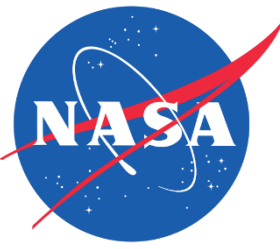


# Li/CF<sub>x</sub>-MnO<sub>2</sub> Temperature Dependence

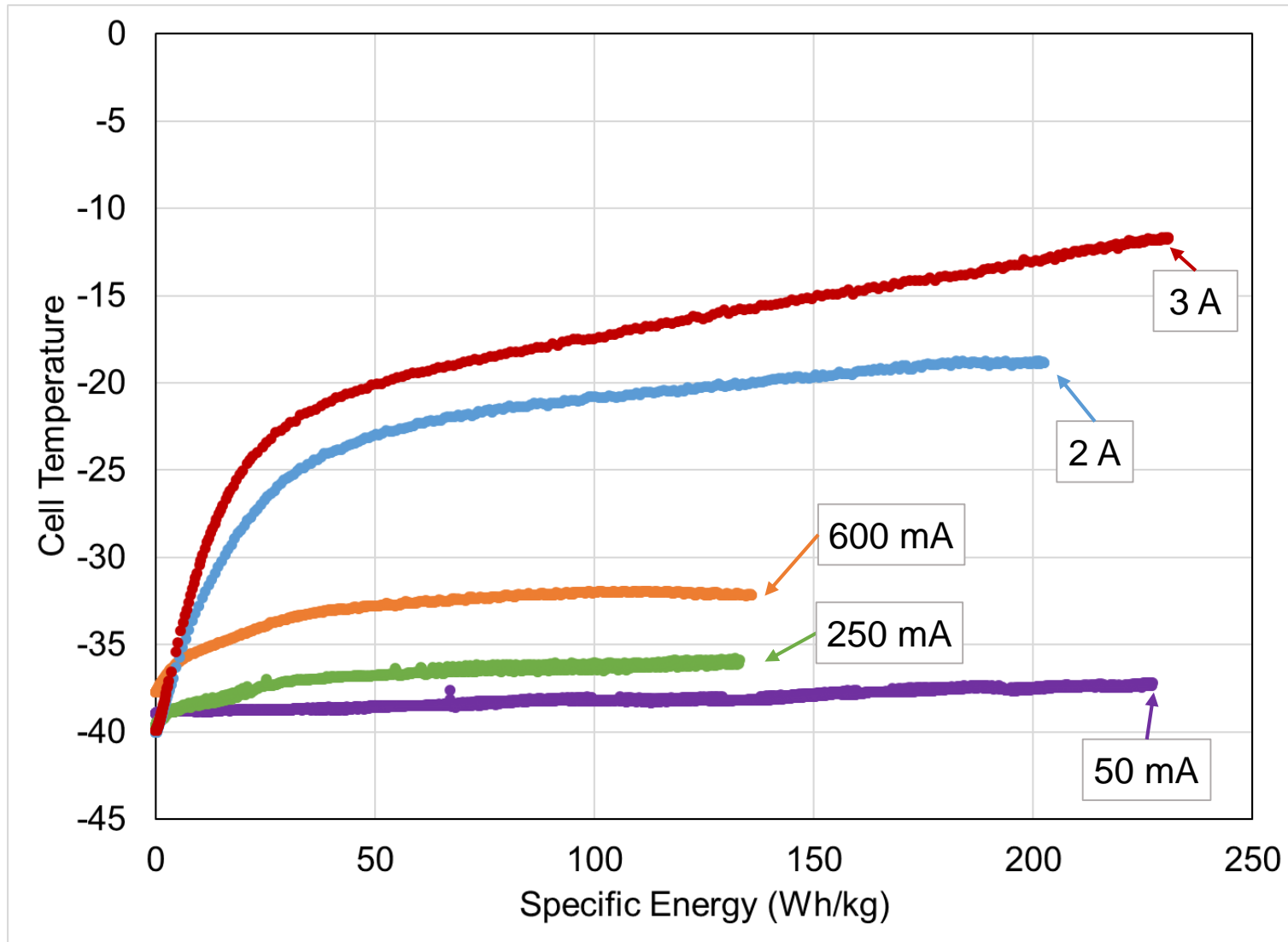


Specific energy falls sharply below -20 °C

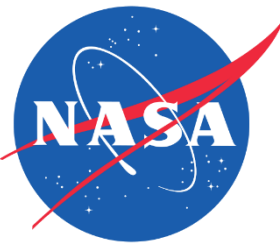




# Li/CF<sub>x</sub>-MnO<sub>2</sub> Discharge Rate at -40 °C

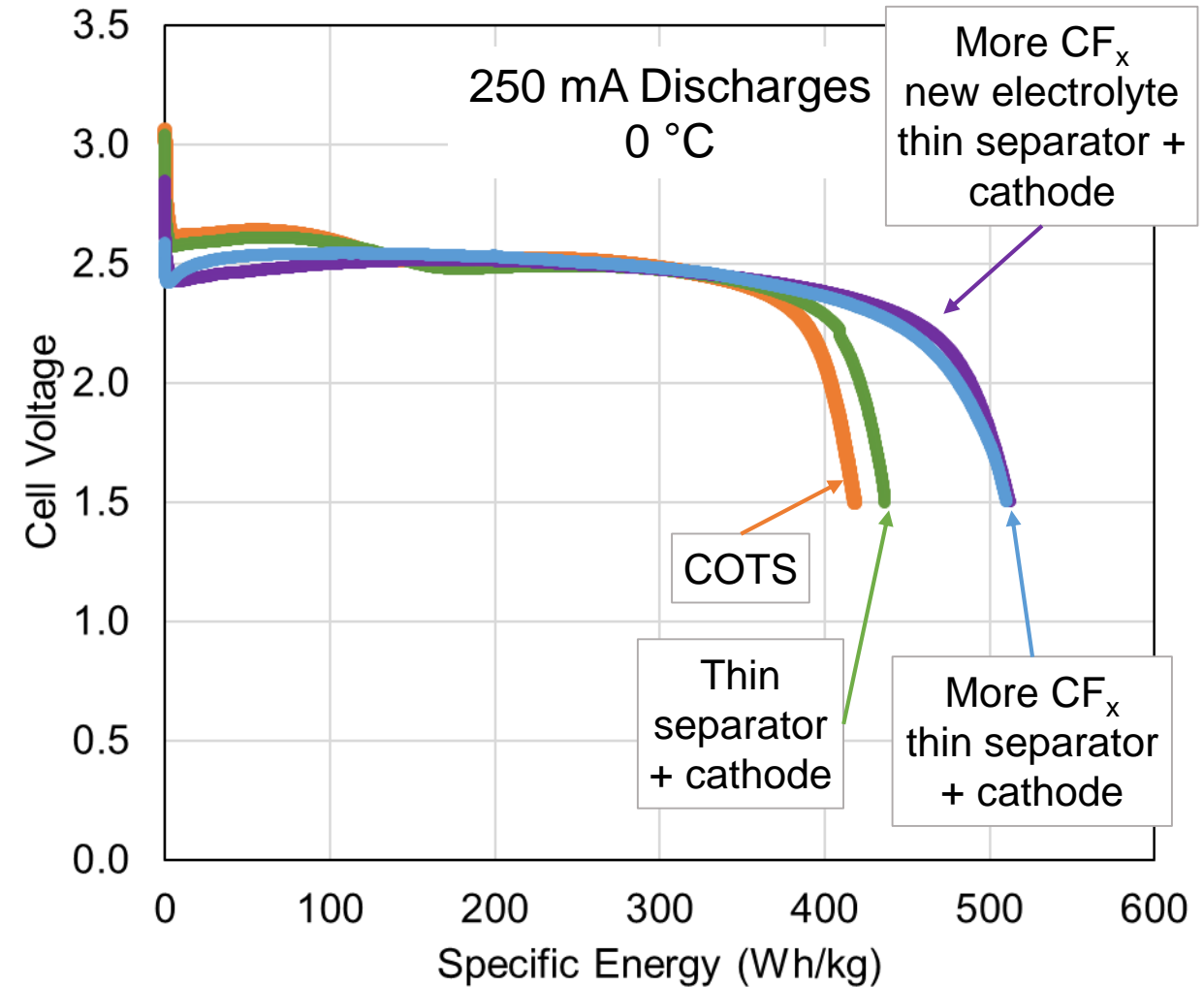
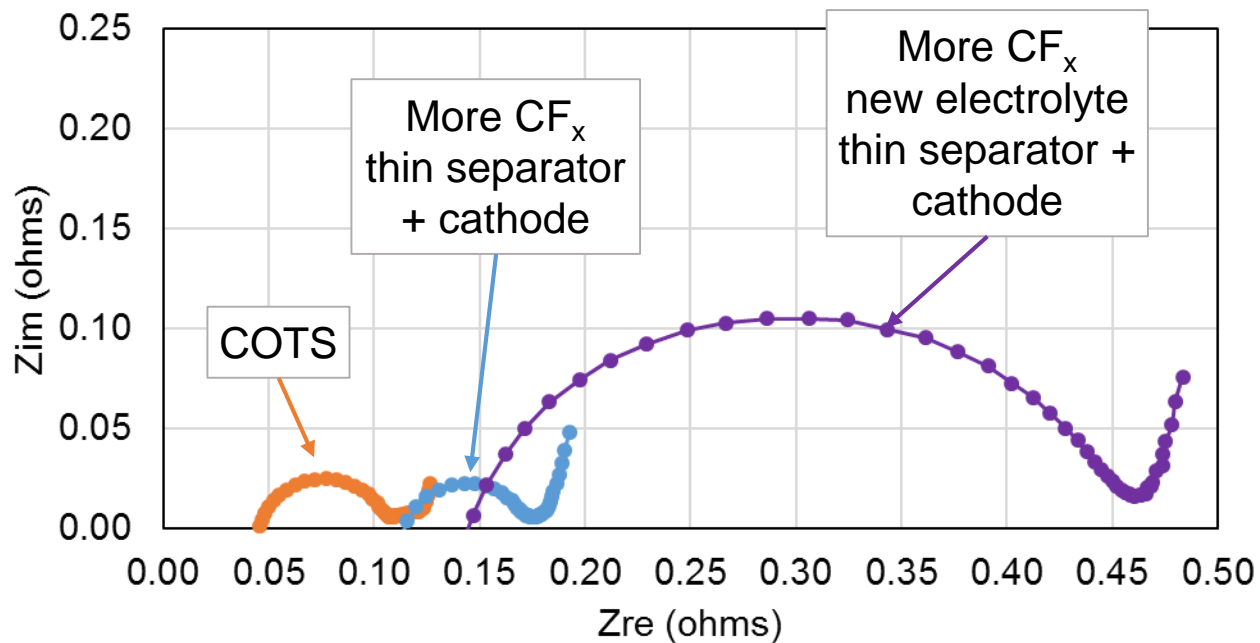


Higher rates lead to more heat

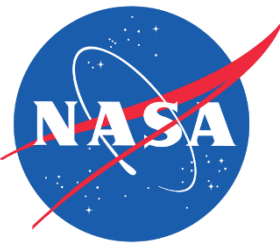


# Eagle-Picher improved cells

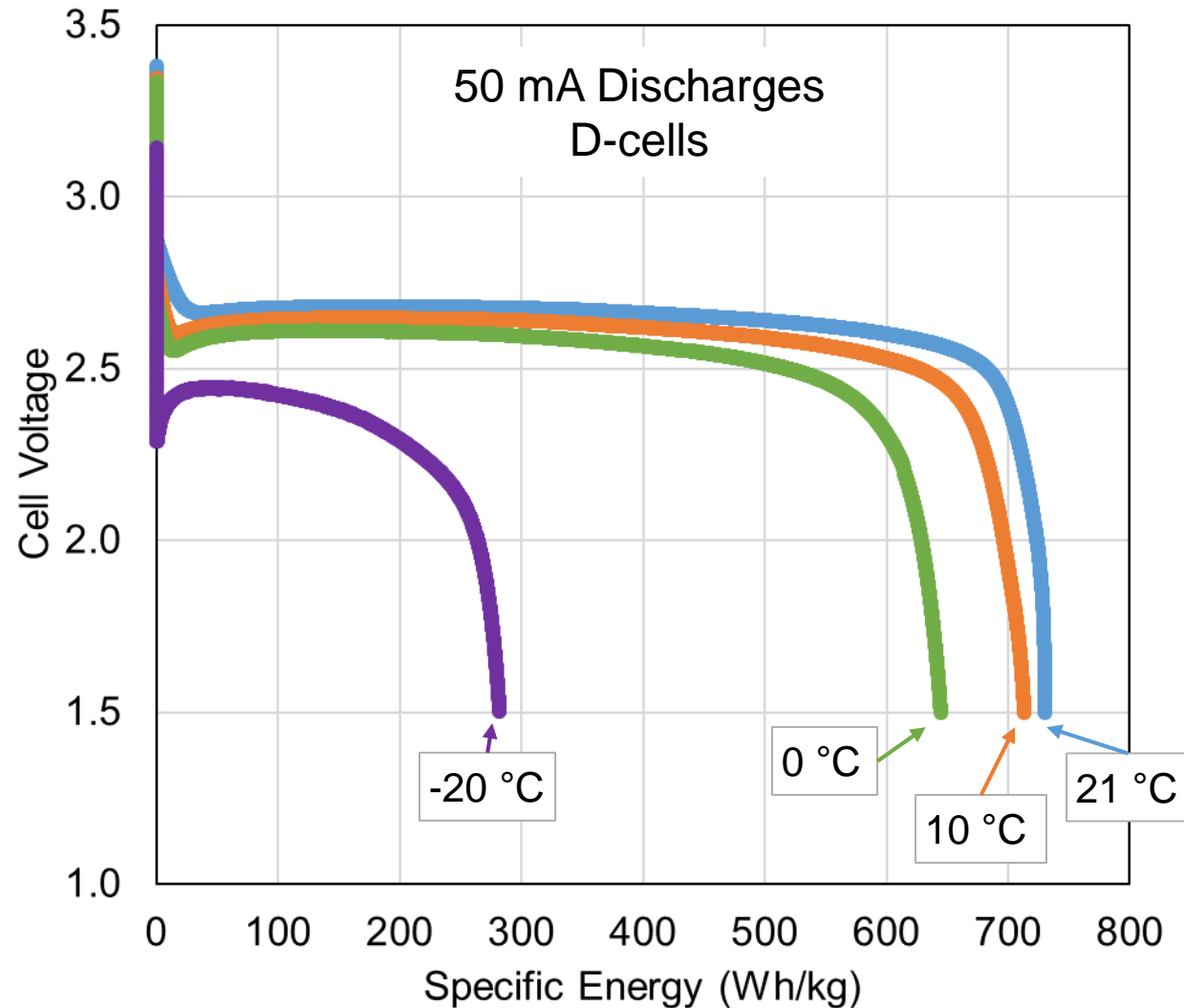
low temperature, low rate targeted



More  $CF_x \rightarrow$  increased cell resistance  
New electrolyte  $\rightarrow$  increased cell resistance



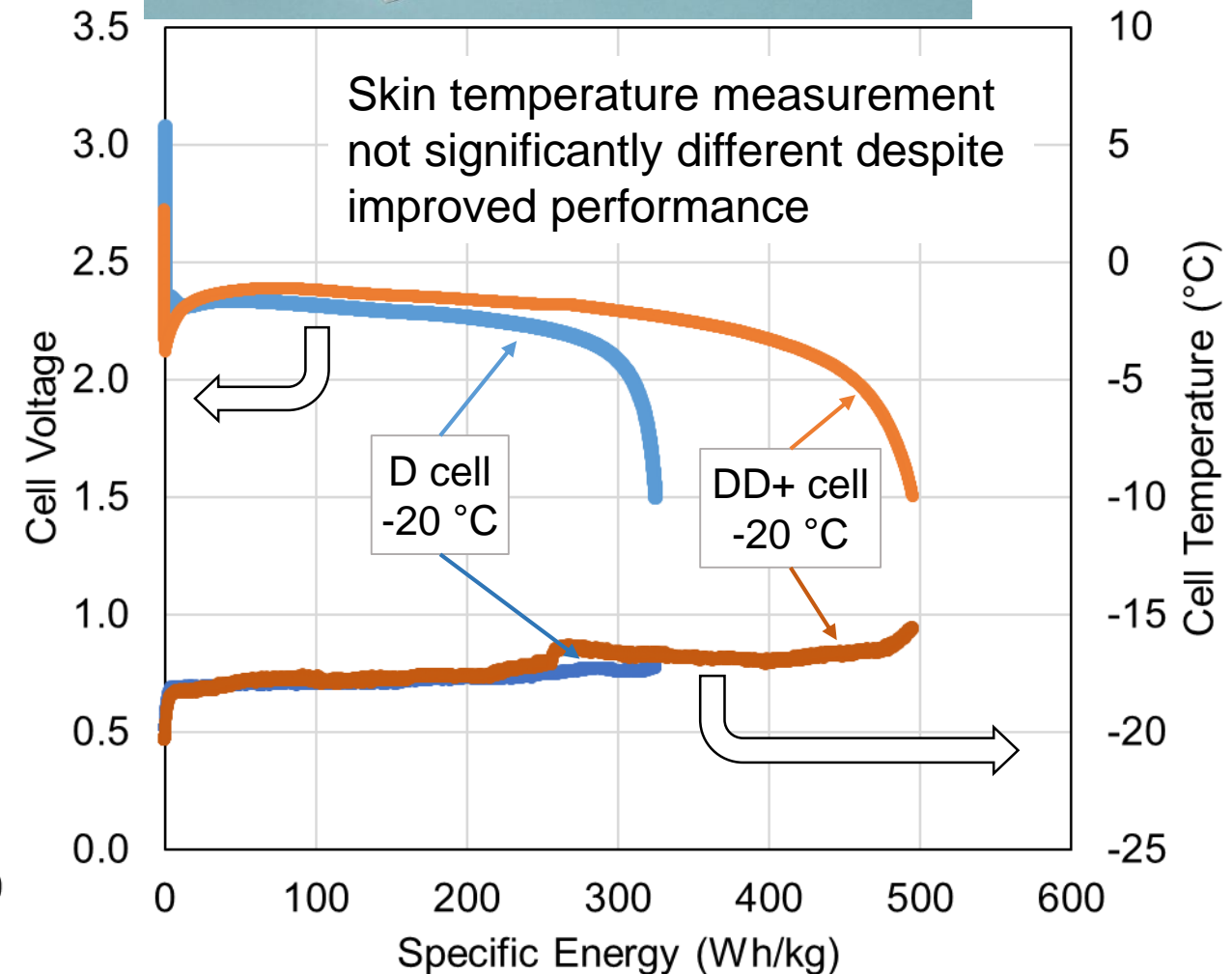
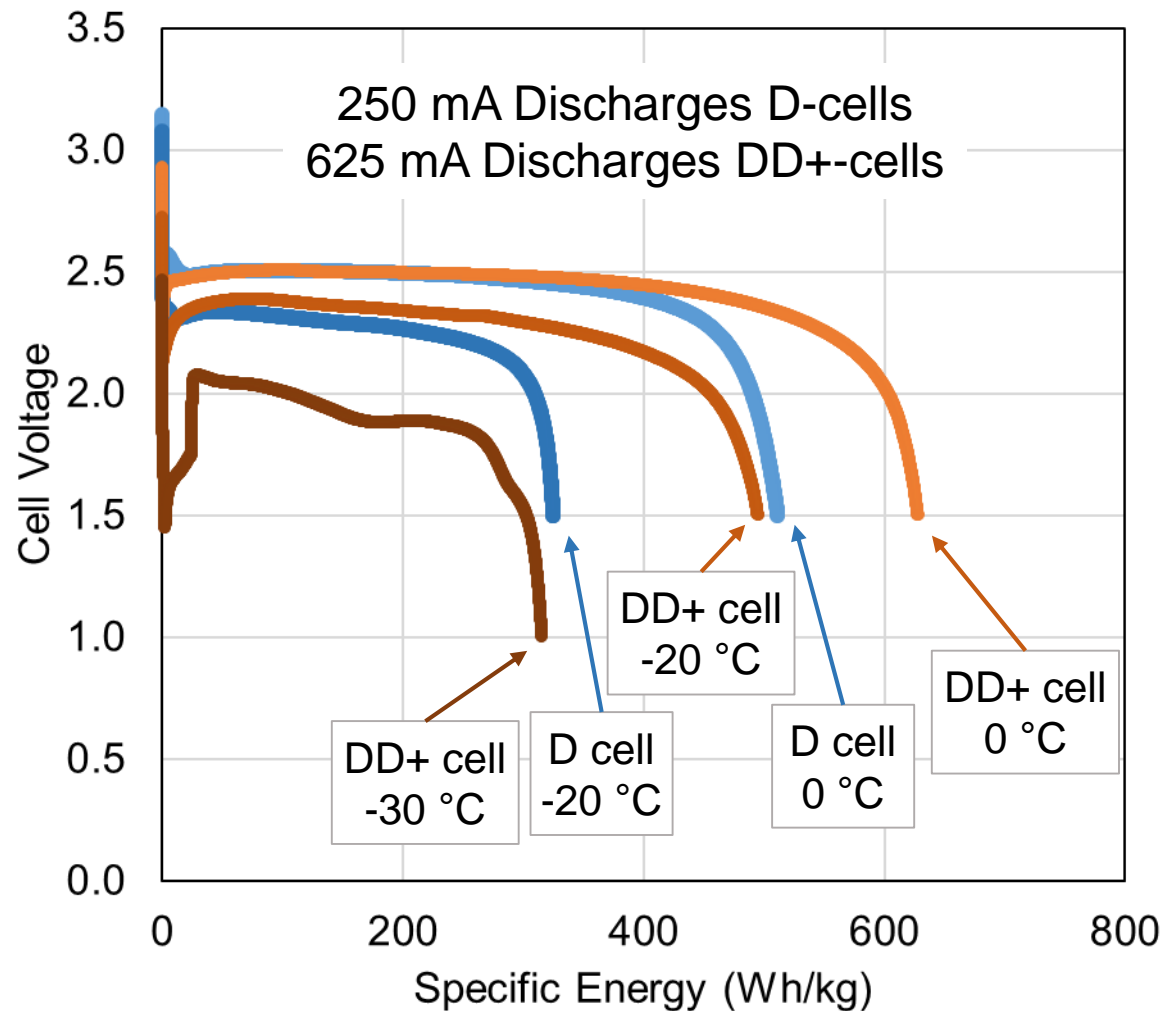
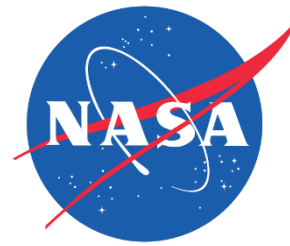
# Li/CF<sub>x</sub> Performance

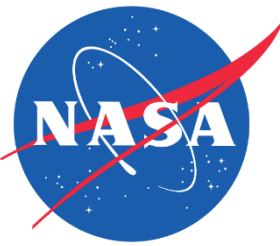


Highest Specific Energy at 50 mA, 21 °C: **730 Wh/kg**

# Li/CF<sub>x</sub> DD+ cell format

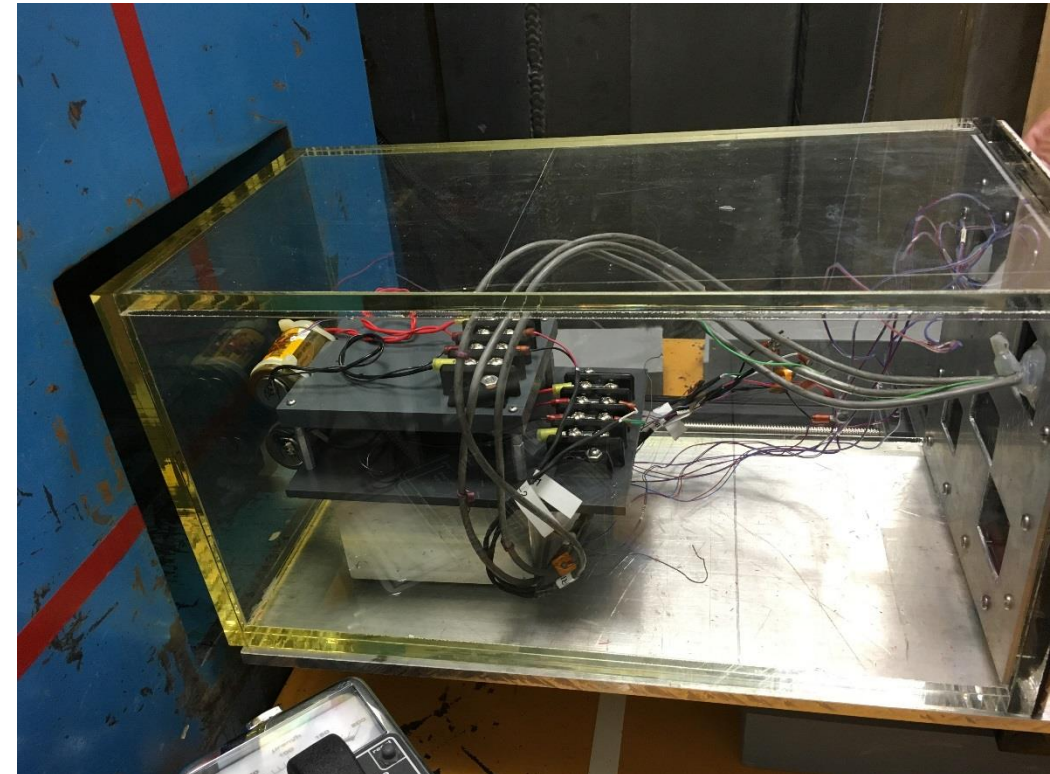
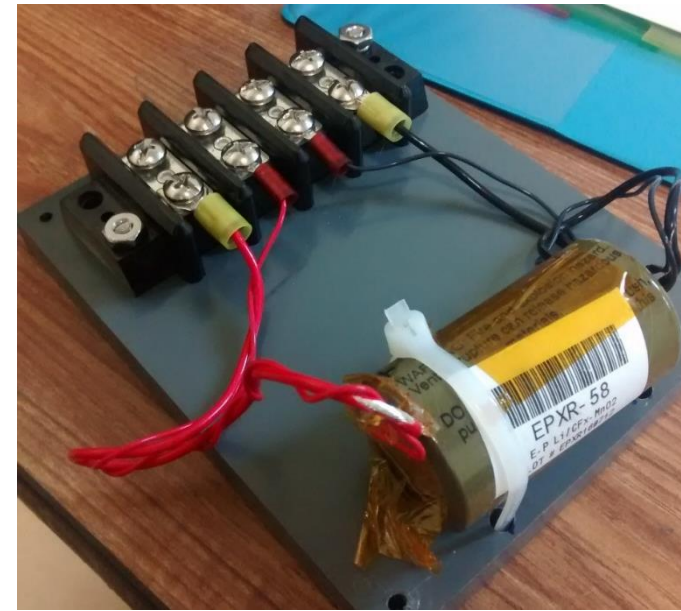
DD+ format improves low temperature performance



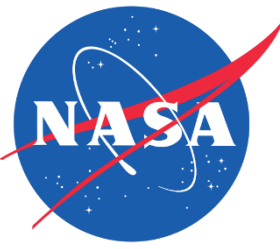


# Radiation Testing

- Long cruise time 6~7 years
- Possible planetary protection protocol
- Eagle-Picher Li/CF<sub>x</sub>-MnO<sub>2</sub> hybrid COTS cells
- JPL high dose rate <sup>60</sup>Co source
  - 1.3 MeV gamma rays
  - ~200 rad/s
  - 1 MRad up to 15 MRad

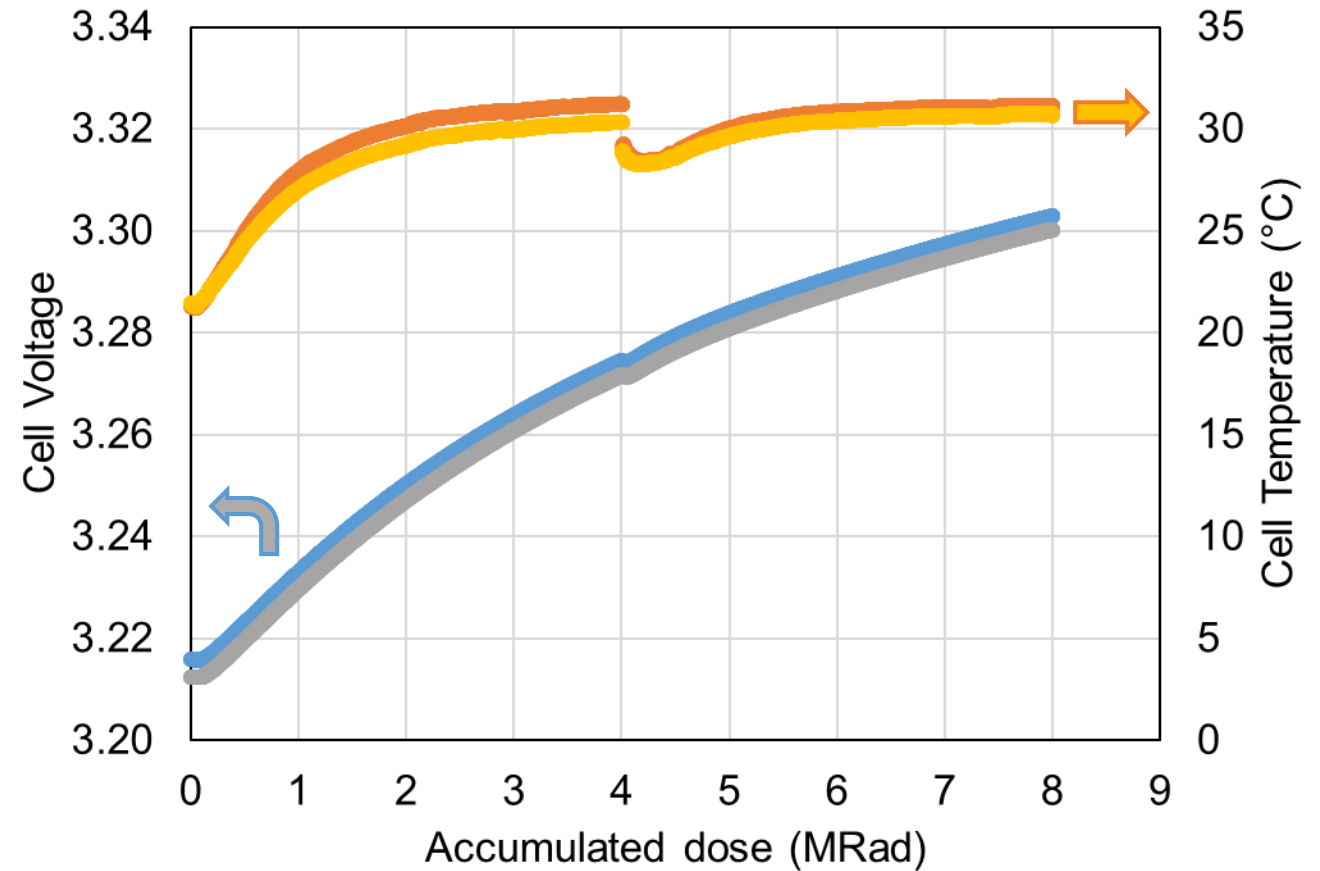
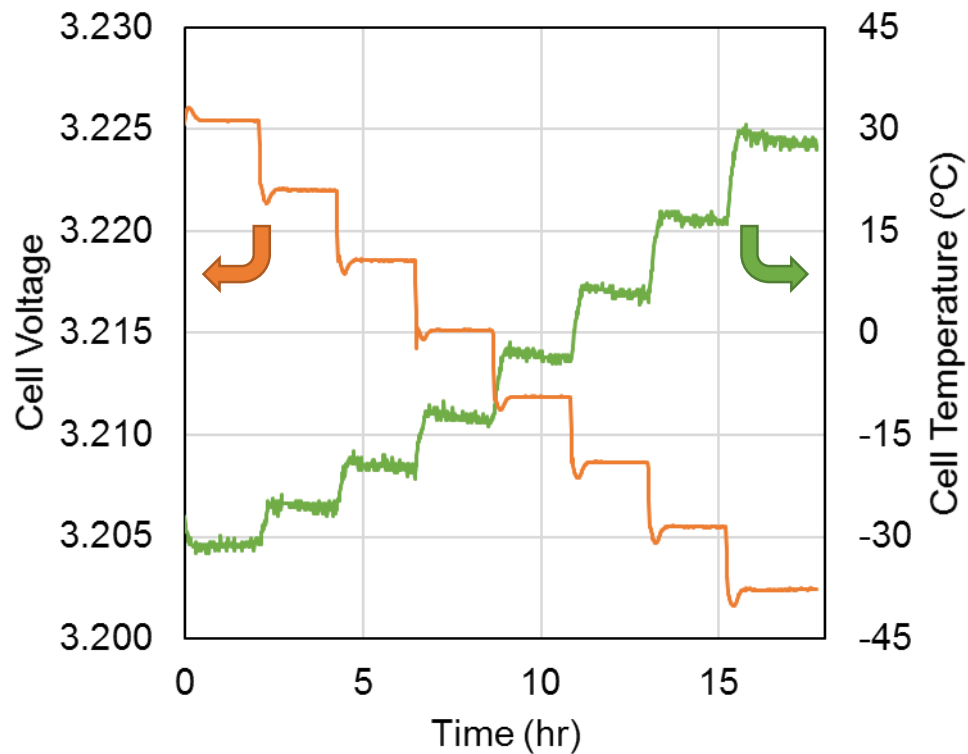


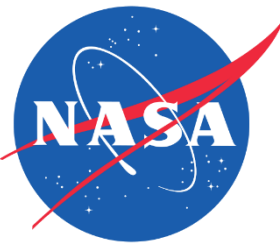




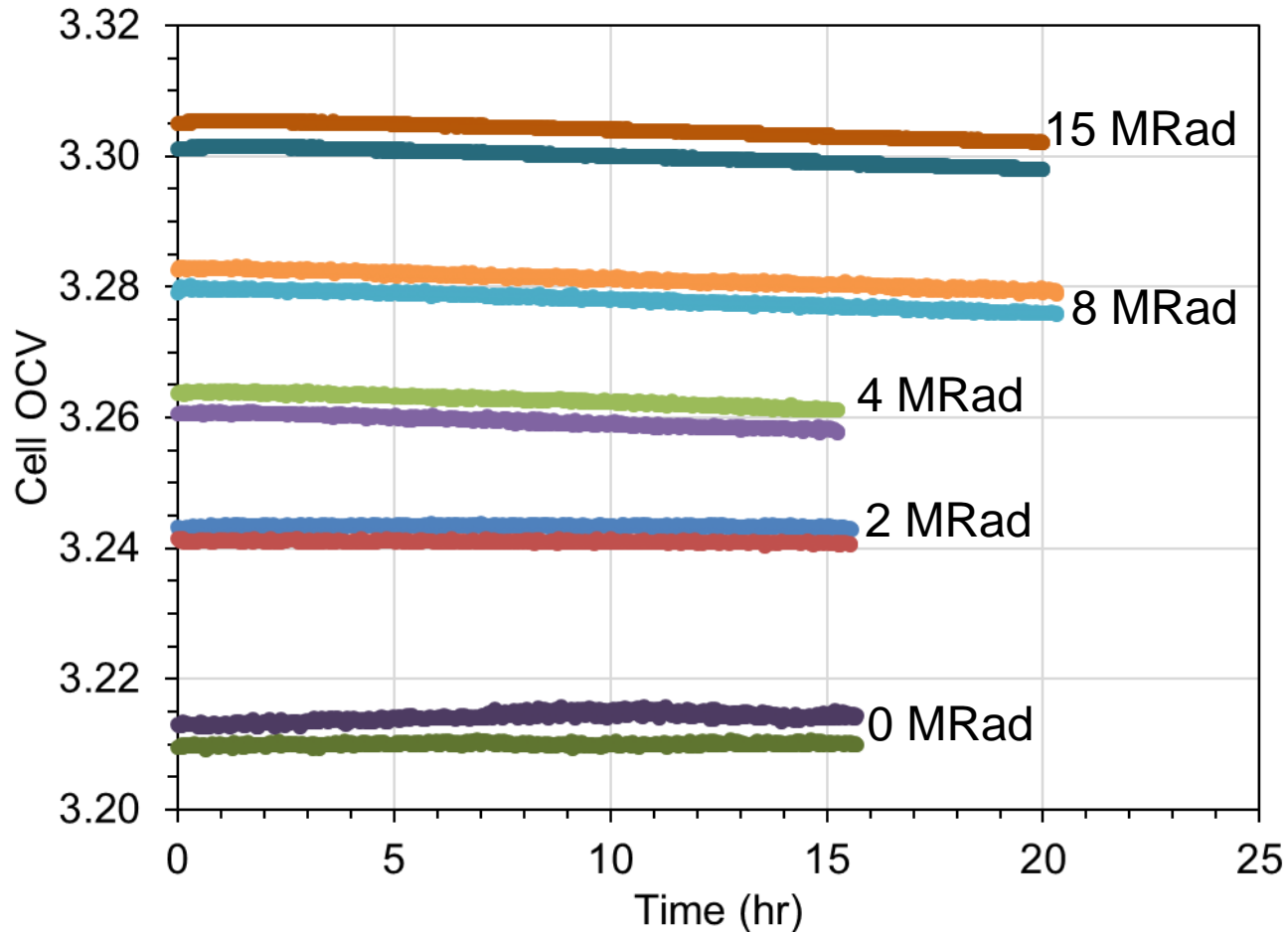
# Li/CF<sub>x</sub>-MnO<sub>2</sub> cell voltages increase during radiation exposure

Temperature rise does not correlate with voltage increase

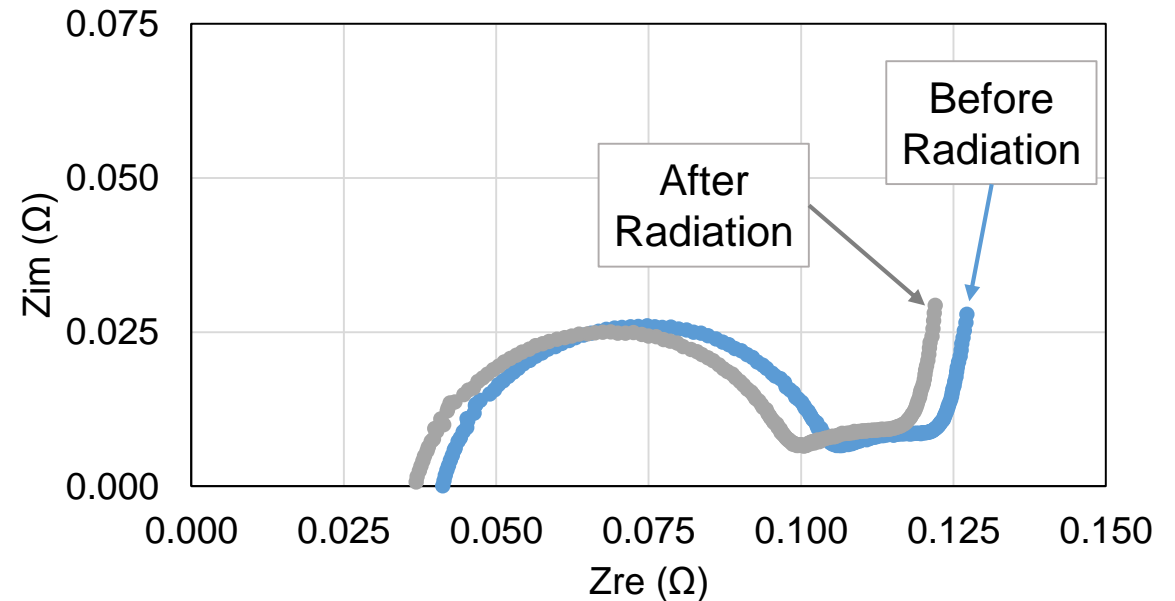




# Voltage Slowly Relaxes Li/CF<sub>x</sub>-MnO<sub>2</sub> Cells



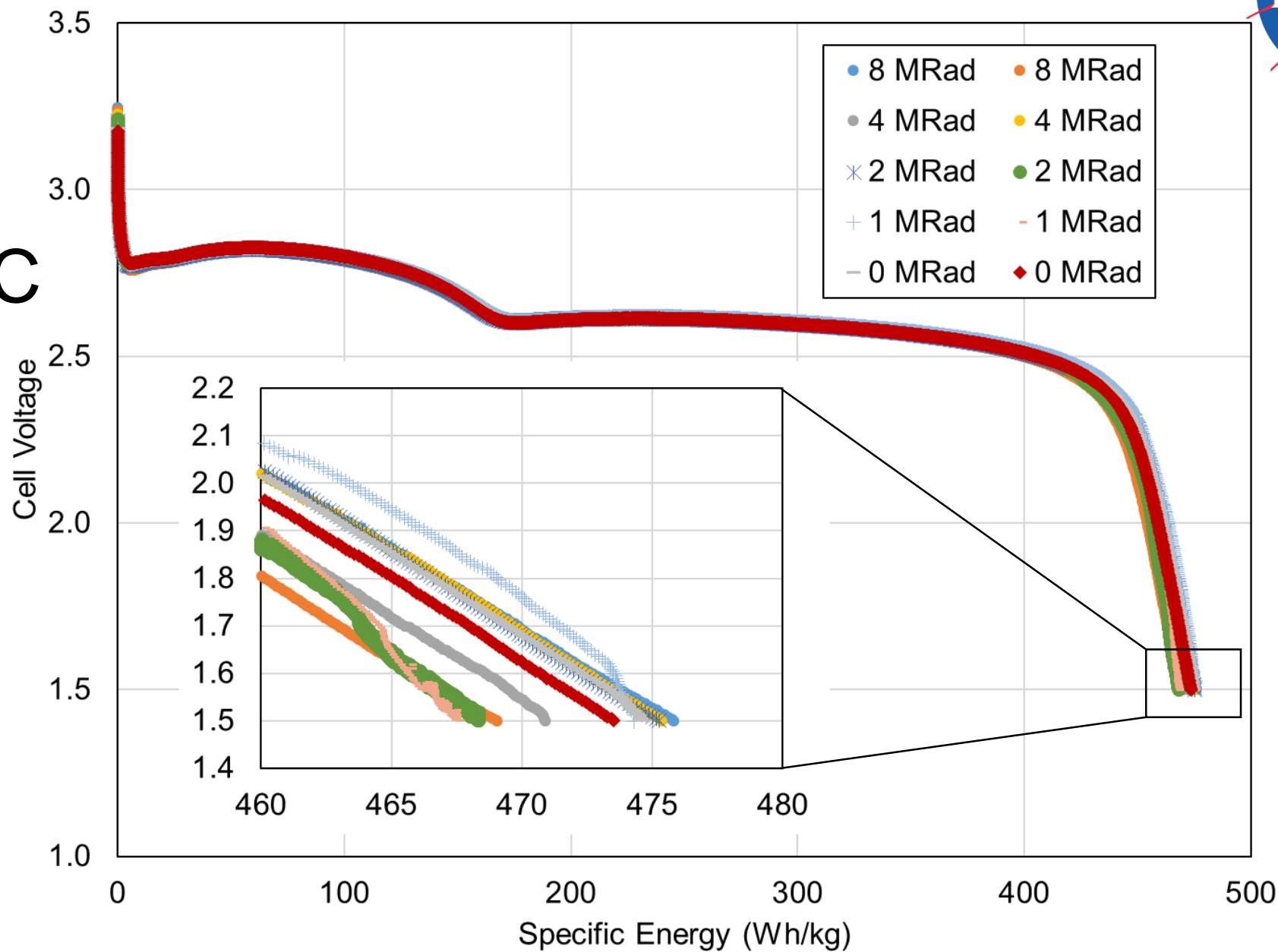
- OCV monitored for >15 hours
- Linear regression analysis shows a slight voltage drop (**0.18 to 0.2 mV/hr**) for higher dose cells (4, 8, 15 MRad)
- No change in impedance after radiation

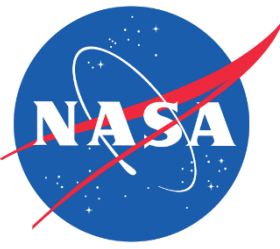




# Li/CF<sub>x</sub>-MnO<sub>2</sub> Radiation cell discharges at 250 mA, 21 °C

- ☒ No negative effects have been observed on specific energy
- ☒ May be useful for planetary protection

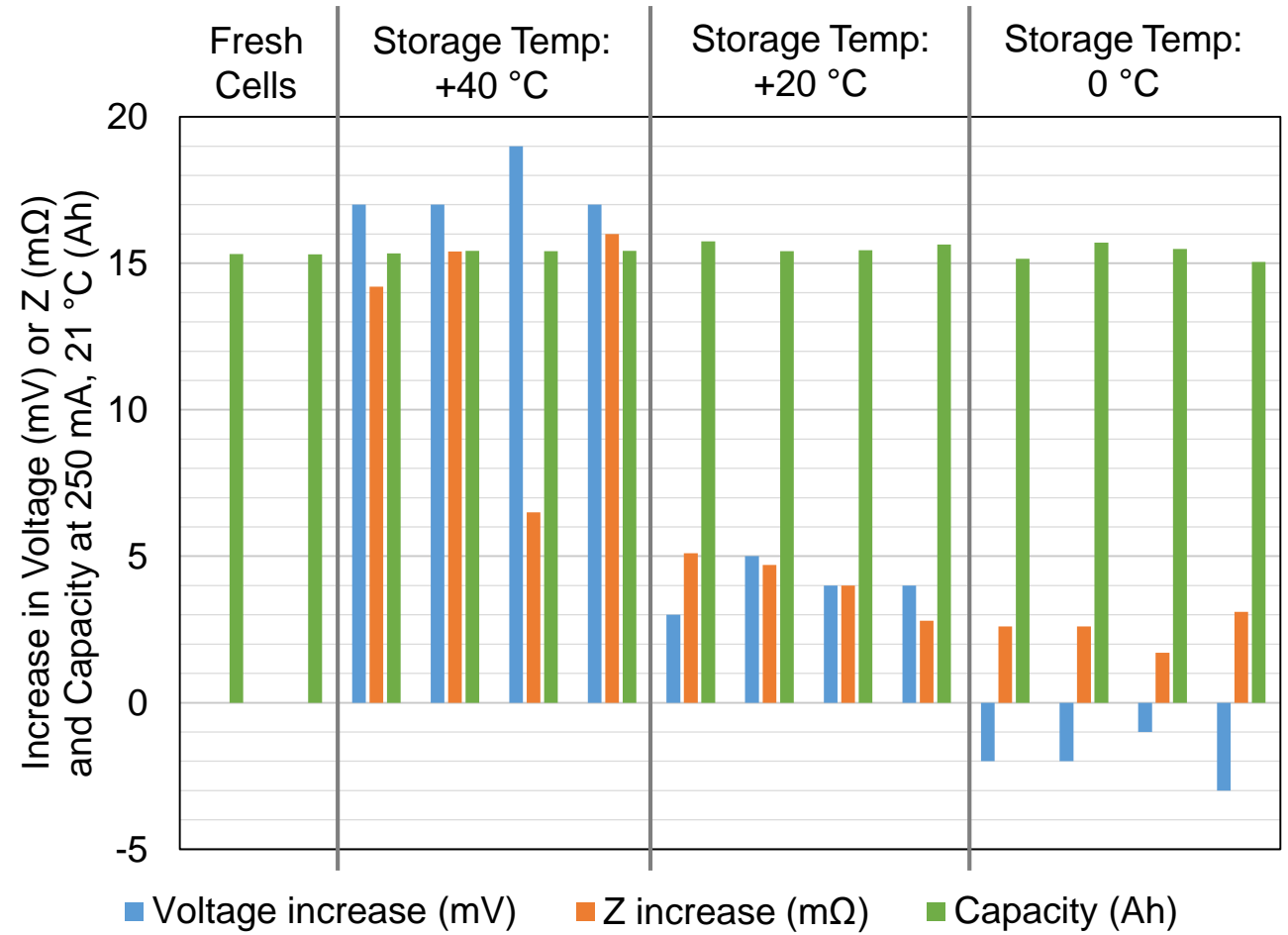


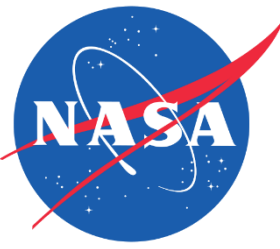


# Storage testing $\text{Li/CF}_x\text{-MnO}_2$

- Storage at 3 temperatures
  - 40 °C
  - 20 °C
  - 0 °C
- Pull cells out after 3 months, then every 6 months for 3 years
- Check impedance, voltage change and discharge performance

## 3 Month Storage Results No impact on capacity

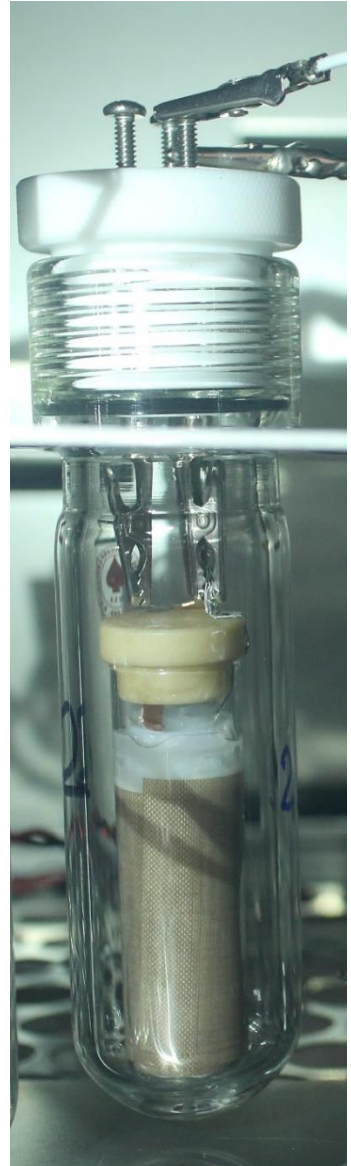
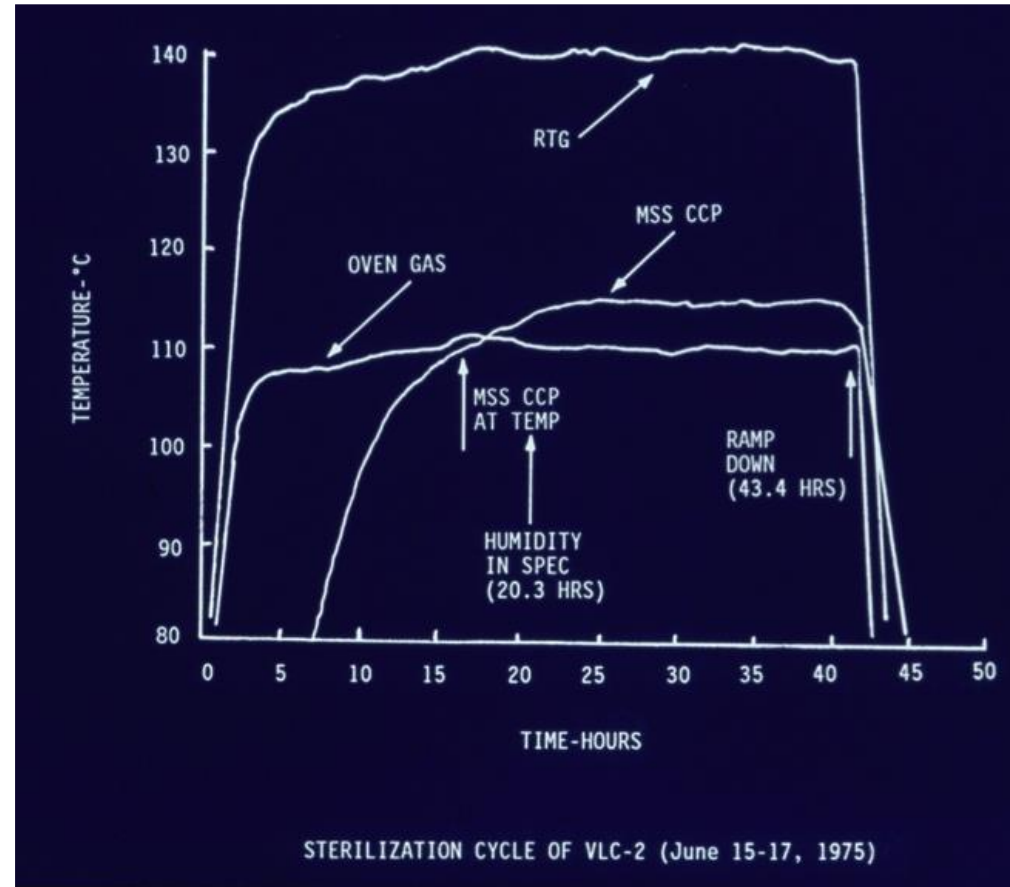




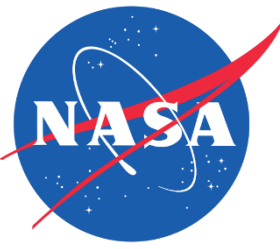
# Experimental Cell Dry Heat Microbial Reduction (DHMR)

- Planetary protection requirements
  - Heat cells to reduce microbial content
- Heat cells to 110-170 °C
  - Minimum of 110 °C based on Viking
  - Lithium melts at 180 °C
  - Cells typically rated to 80 °C
- Discharge to understand effects
  - Room temp discharge to measure capacity
  - Low temperature to measure increased resistance

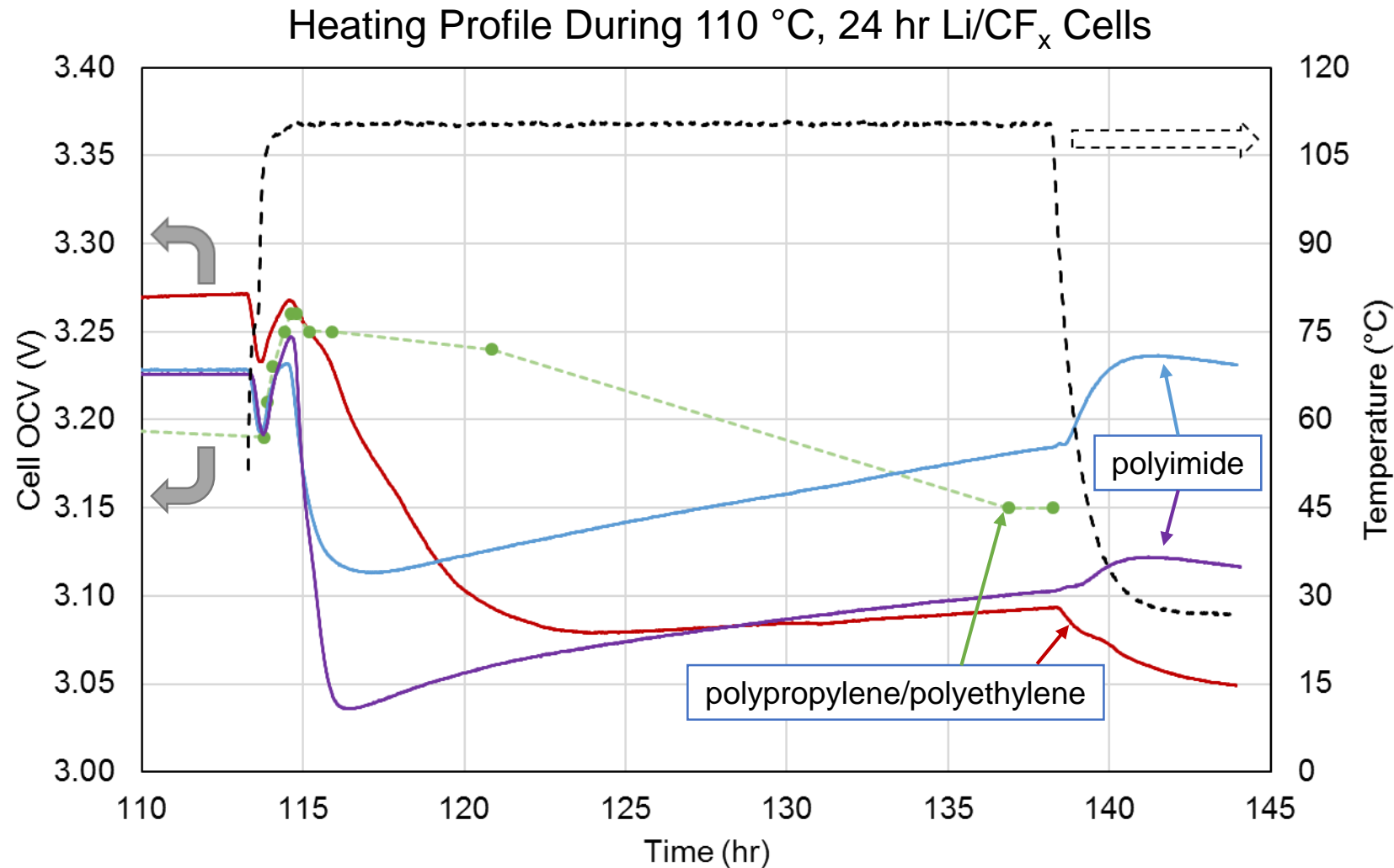
Viking DHMR



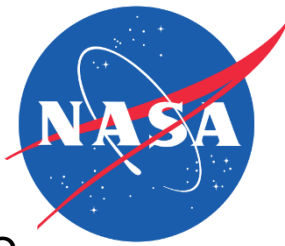




# Dry Heat Microbial Reduction (DHMR) Evaluation for Planetary Protection

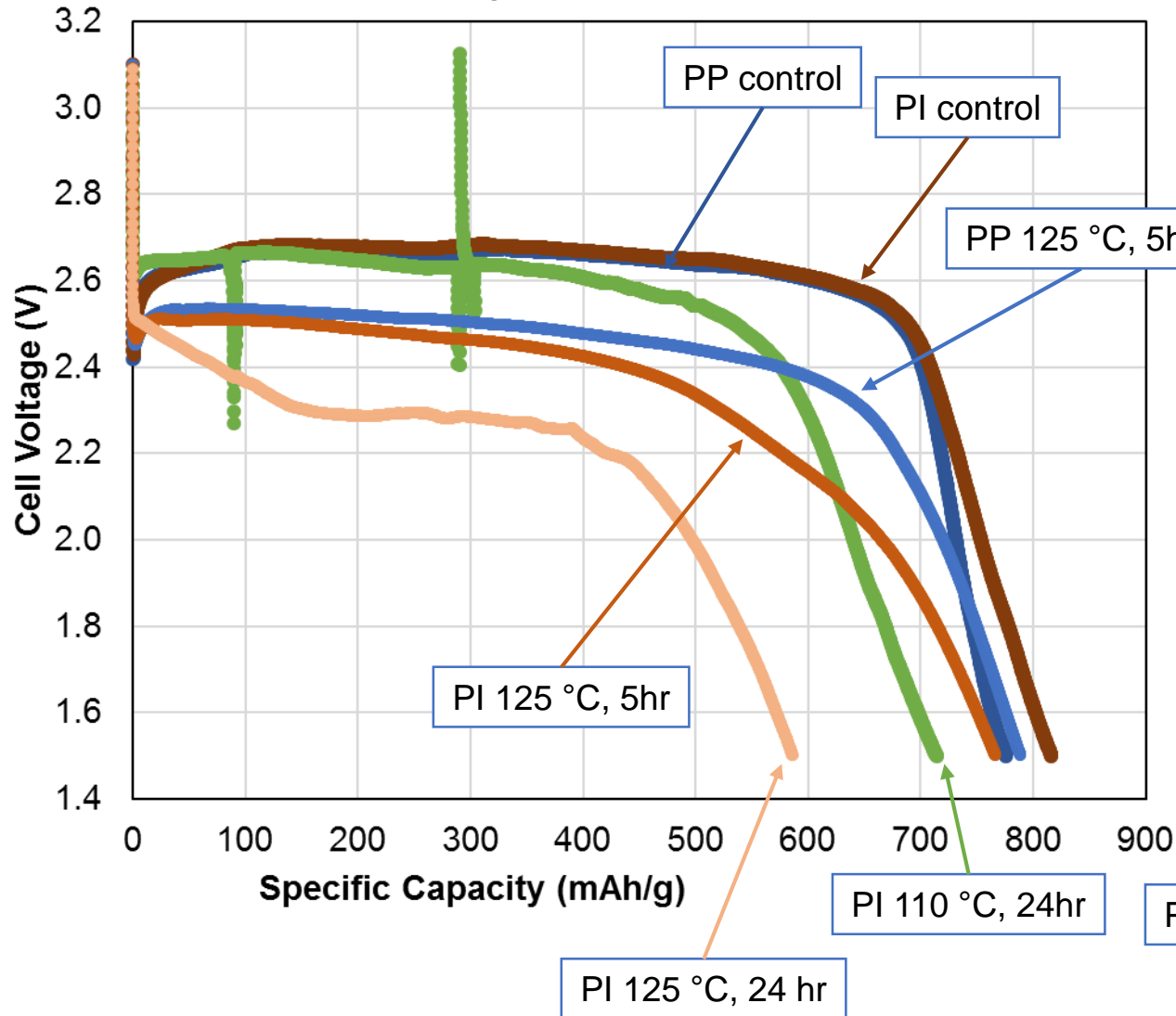


# Discharge Performance After DHMR

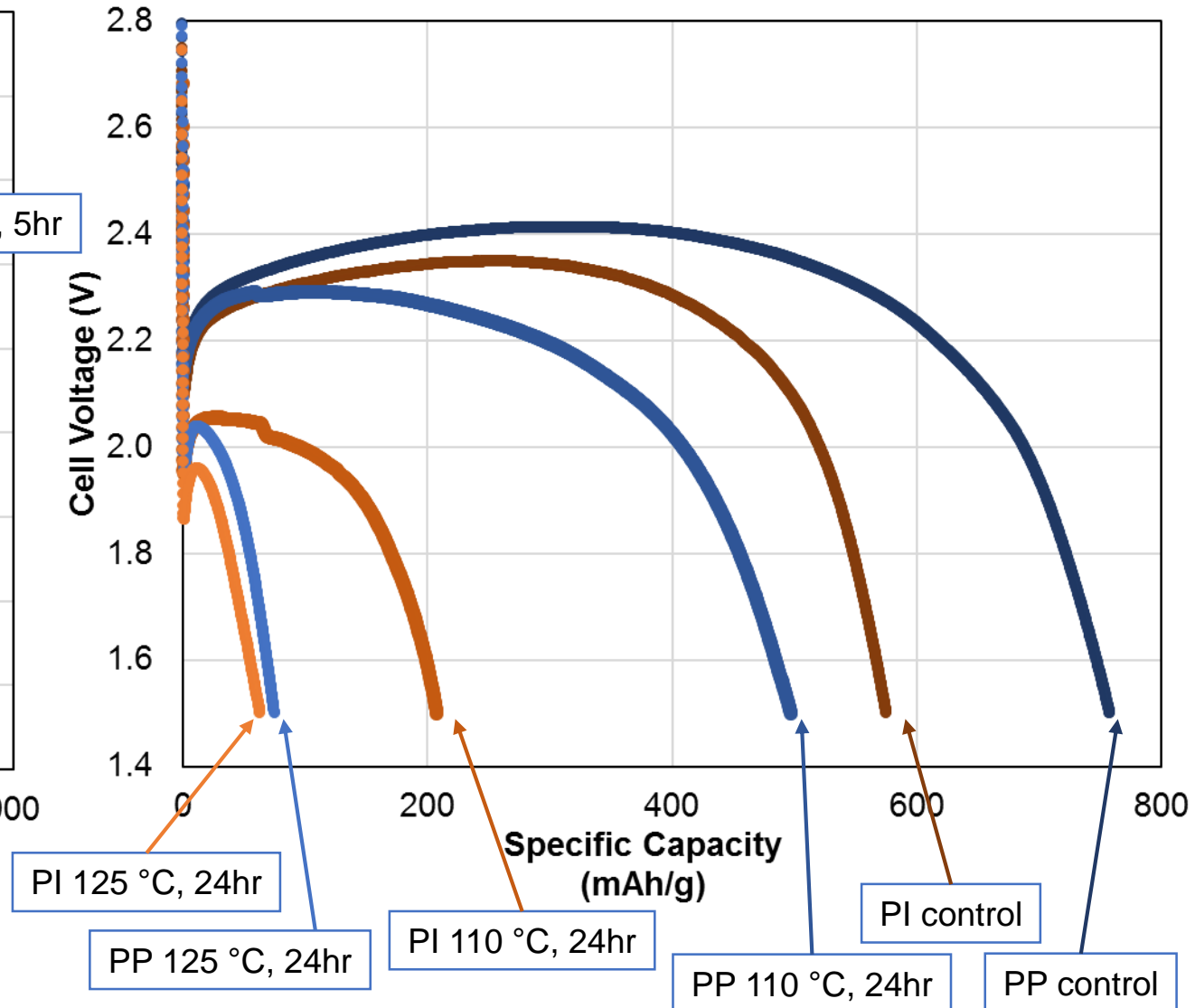


☒ DHMR does not look promising for Li/CF<sub>x</sub> cells

Discharge (4.1 mA) at Room Temperature



Discharge (4.1 mA) at -20 °C



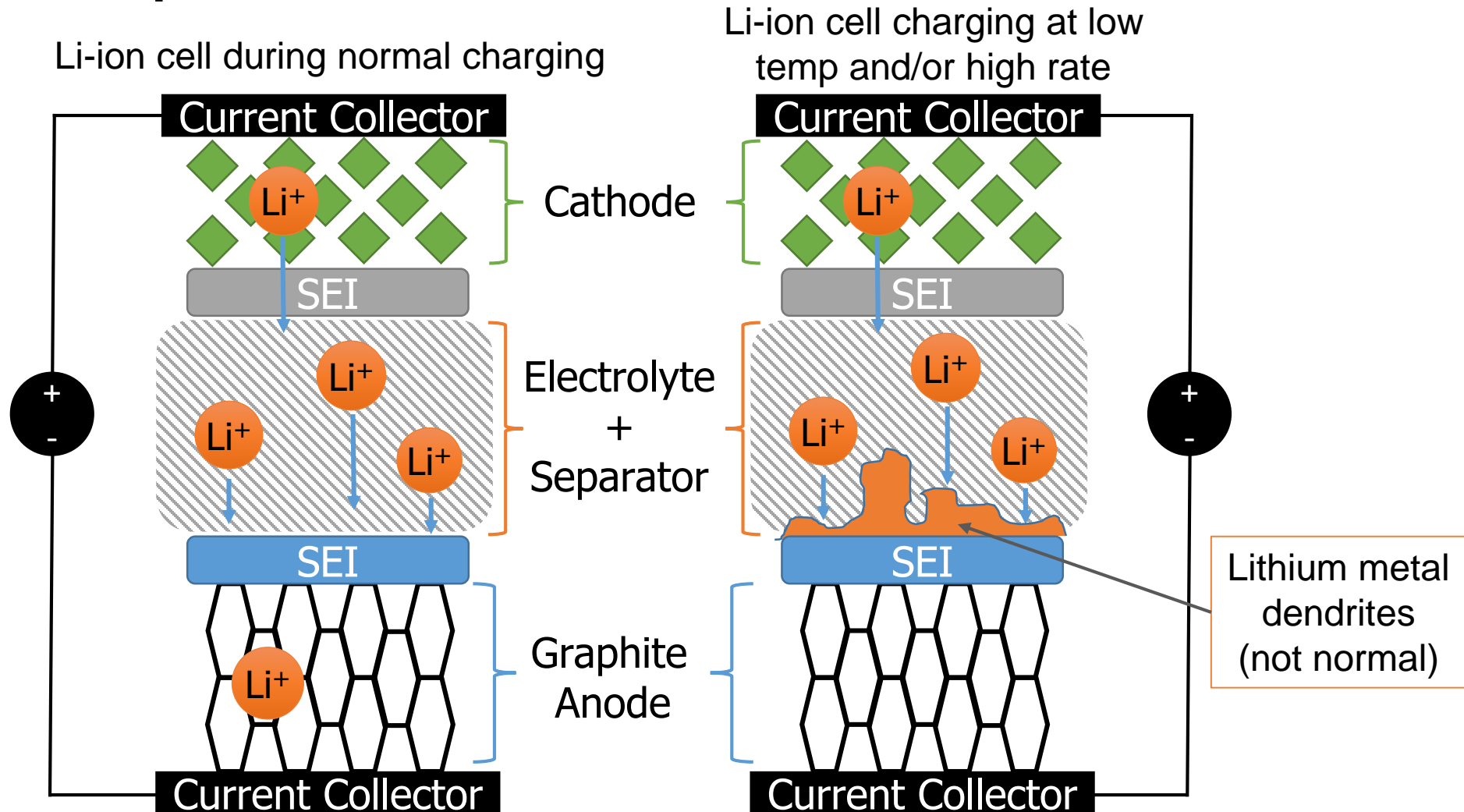
# Lithium Primary Battery Conclusions

- High specific energy (730 Wh/kg) available from Li/CF<sub>x</sub> cells
  - >2X more than heritage Li/SOCl<sub>2</sub> cells
  - Low temperature/high rate performance is poor
- Moderate specific energy (550 Wh/kg) available from improved Li/CF<sub>x</sub>-MnO<sub>2</sub> cells
  - ~50 % higher than heritage Li/SOCl<sub>2</sub> cells
  - Still function at low temperature/high rate
- High dose radiation (up to 10 MRad) does not appear to significantly affect specific energy
  - Cell voltage is impacted
  - Impedance is not impacted
  - May be useful for planetary protection
- DHMR unlikely to be a viable option for Li/CF<sub>x</sub> cells

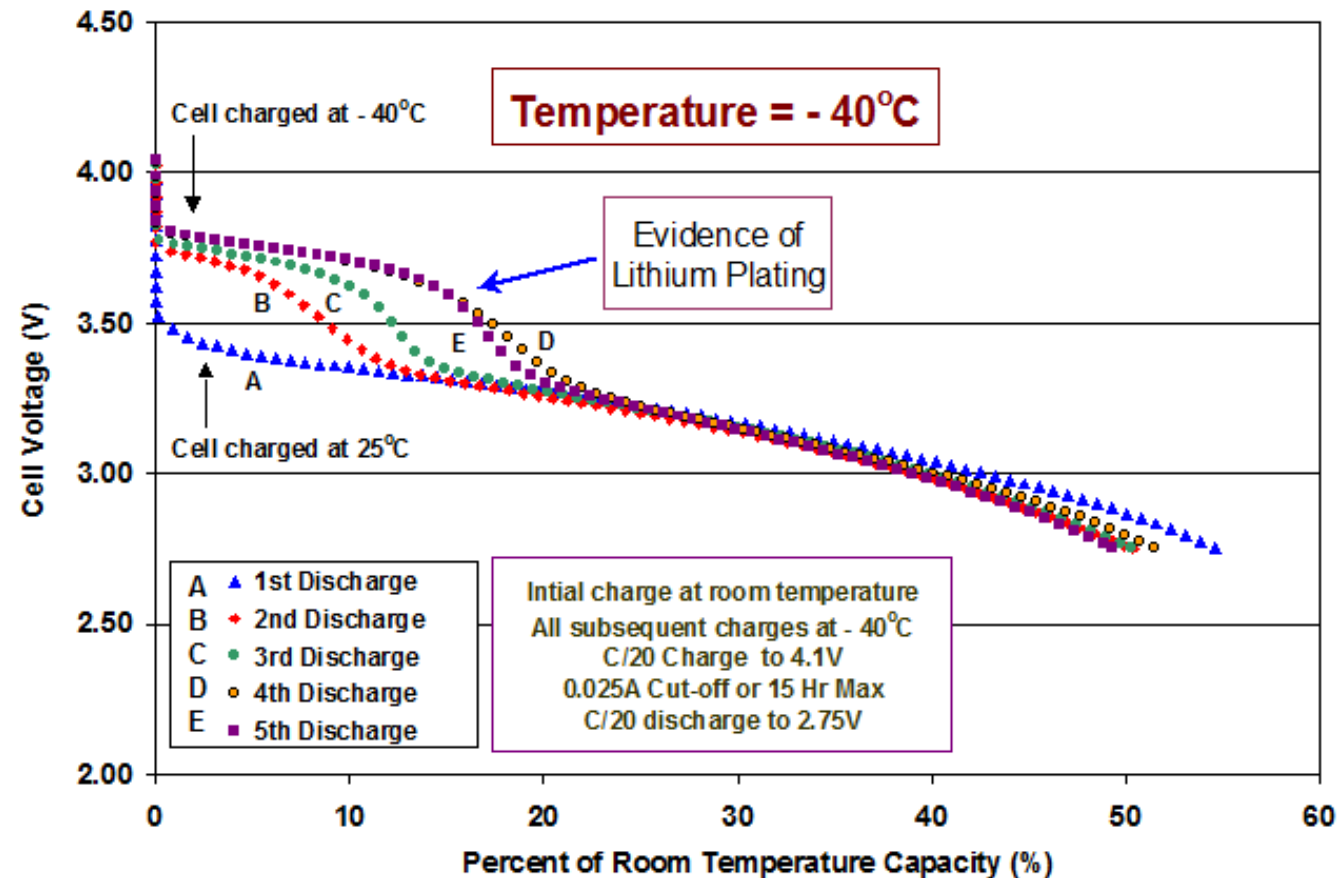
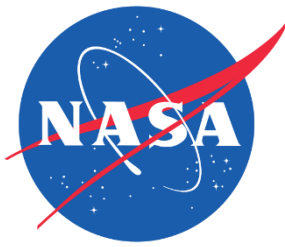


Li/CF<sub>x</sub> cell after heating to 170 °C

# Secondary (Rechargeable) Batteries for Low Temperatures

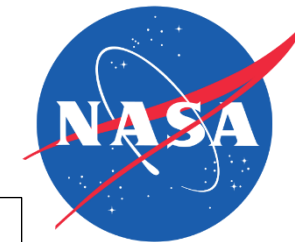


# Previous work at JPL to detect and quantify plating



- (1) M. C. Smart, B. V. Ratnakumar, L. Whitcanack, K. Chin, M. Rodriguez, and S. Surampudi, *IEEE Aerospace and Electronic Systems Magazine*, 17 (12), 16-20 (2002).
- (2) M. C. Smart and B. V. Ratnakumar, *J. Electrochem. Soc.*, **158** (4), A379 (2011).

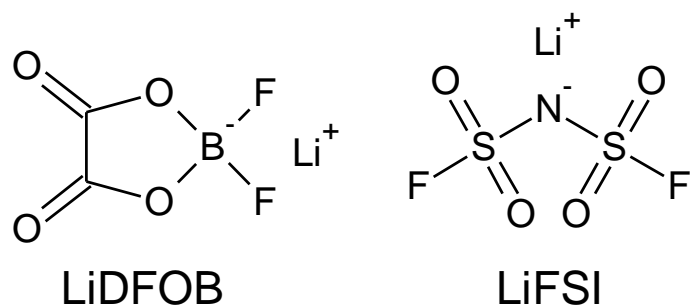
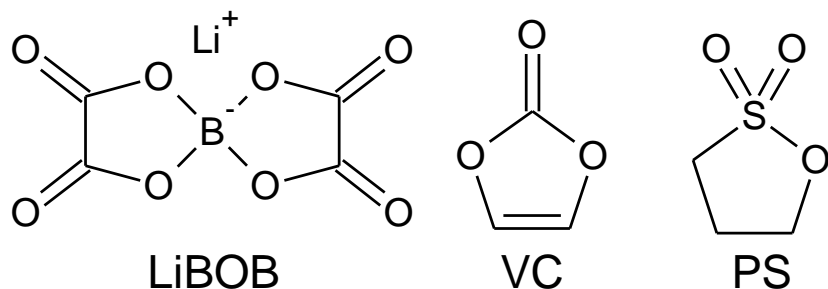




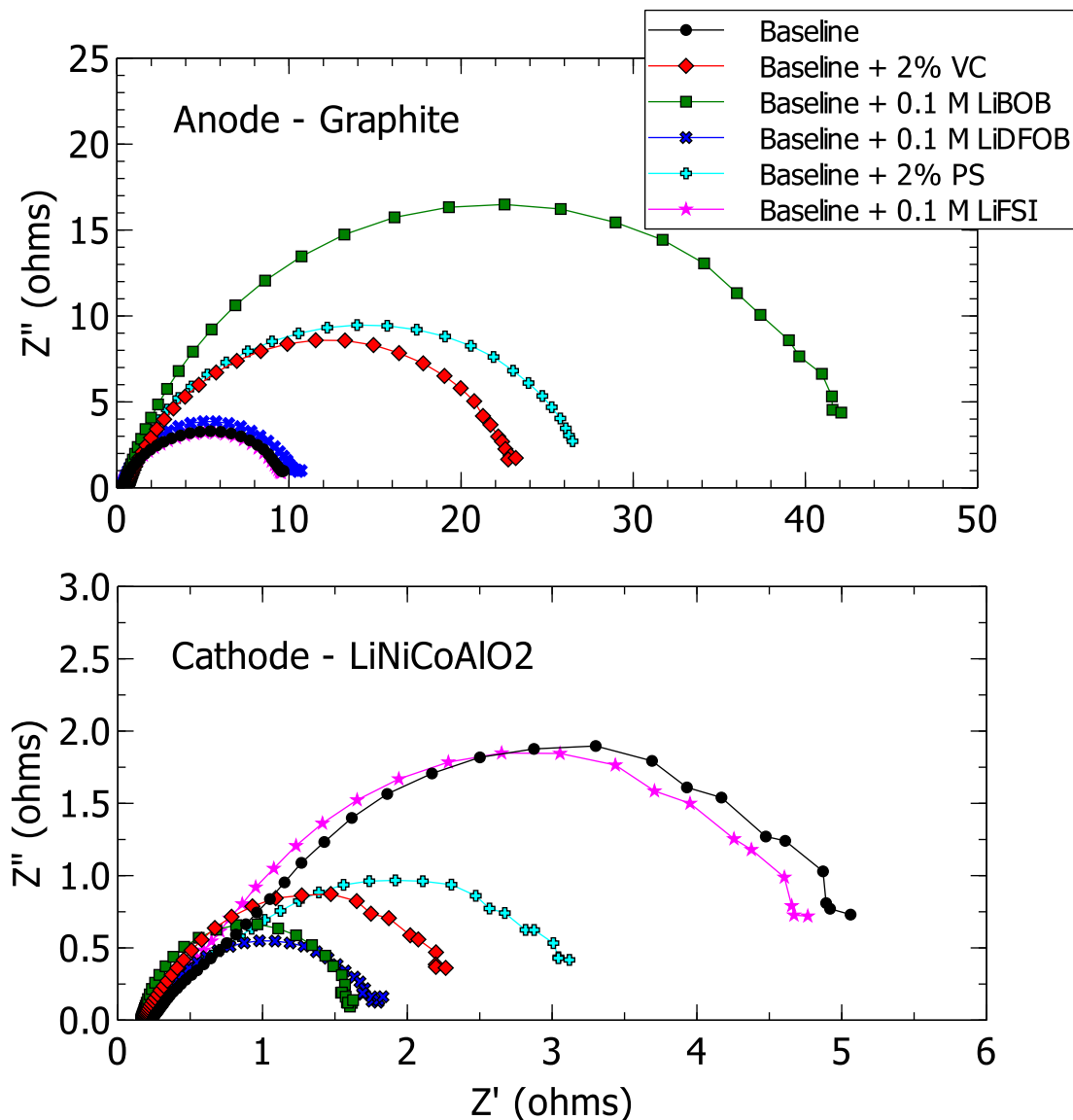
# Impedance for Graphite/LiNiCoAlO<sub>2</sub> Cells at -30 °C

Baseline Electrolyte: 1.0 M LiPF<sub>6</sub> in  
EC+EMC+MP (20:20:60 vol. %)

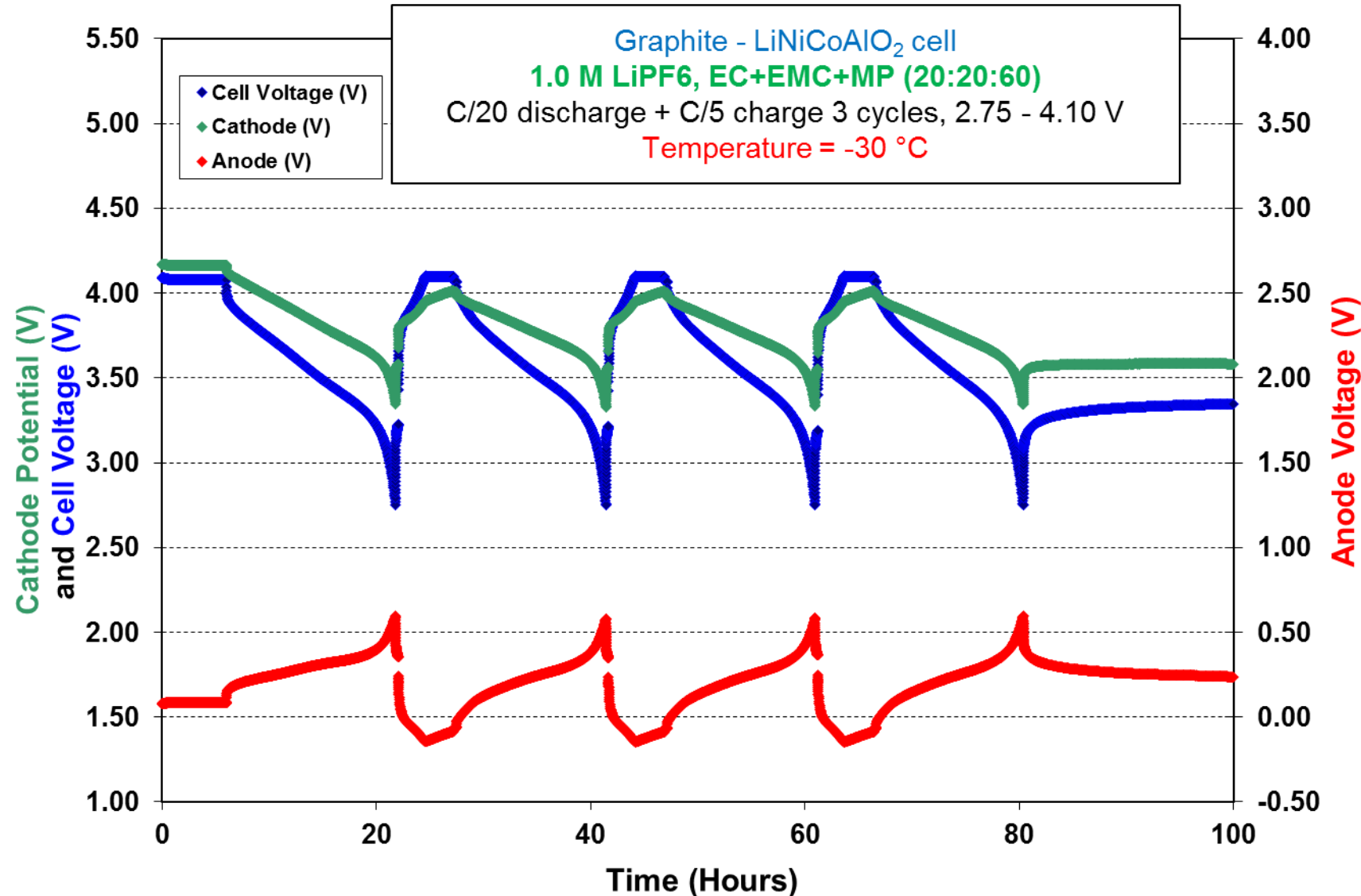
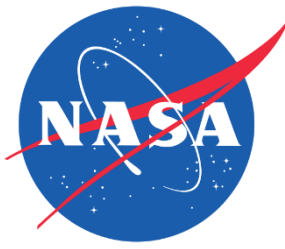
## Additives Investigated



Anodes are more resistive than cathodes overall and additives other than LiFSI decrease cathode resistance and increase anode resistance



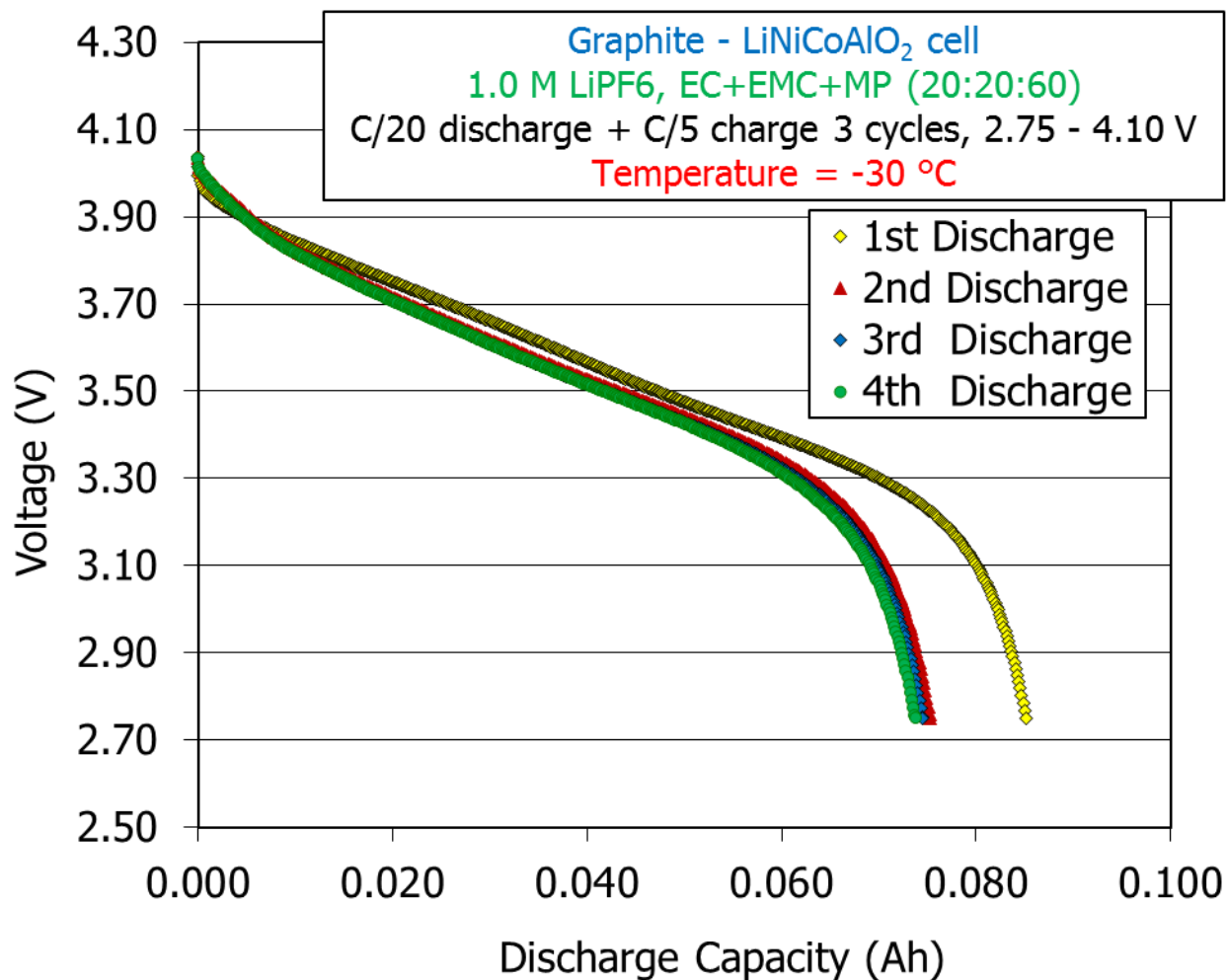
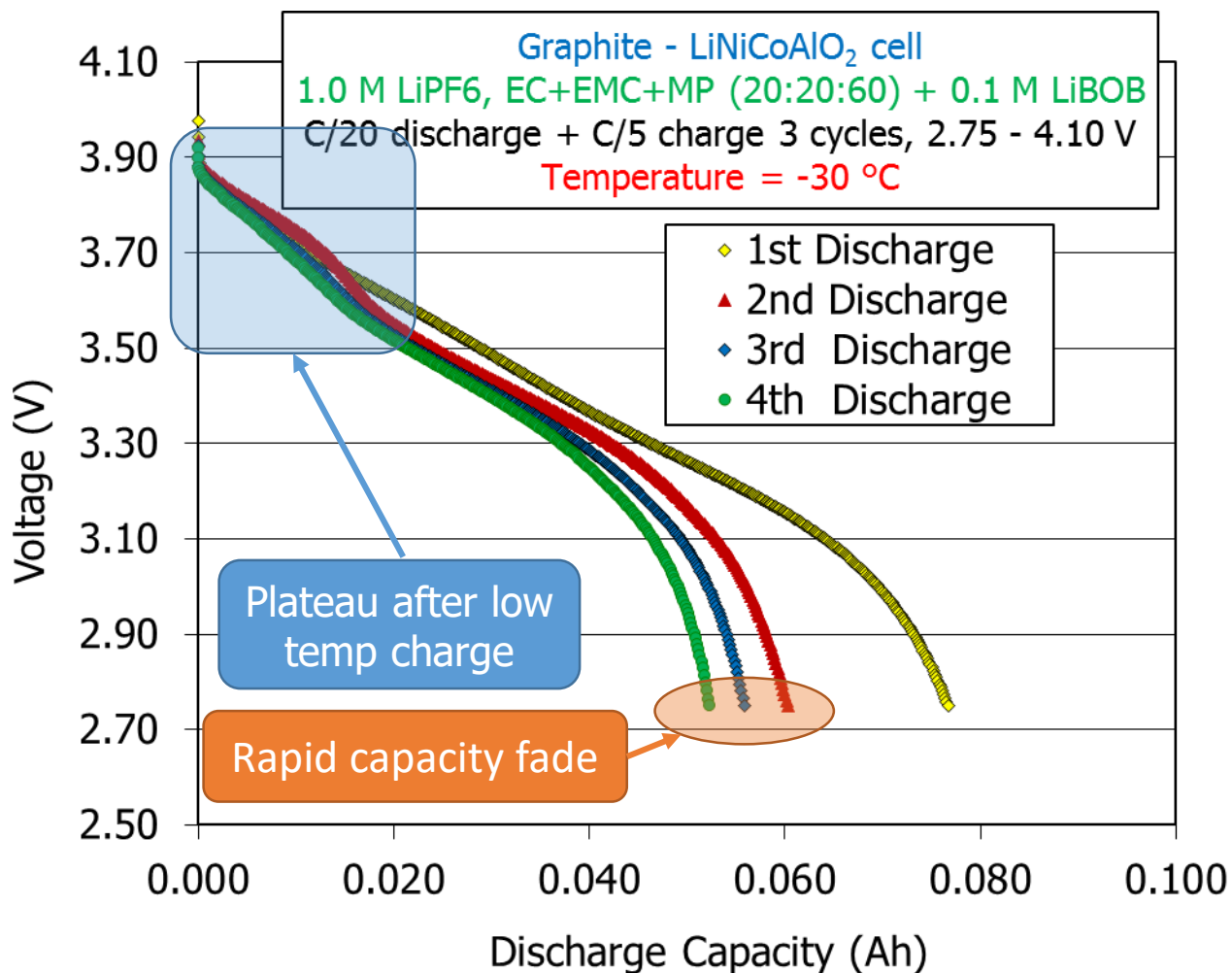
# Testing Procedure to Detect Lithium Plating - Baseline

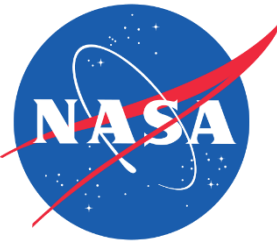


1. Charge at room temp
2. Discharge at C/20 at low temp
3. Charge at C/5 with C/50 taper at low temp
4. Discharge at C/20
5. Repeat steps 3 and 4 two more times



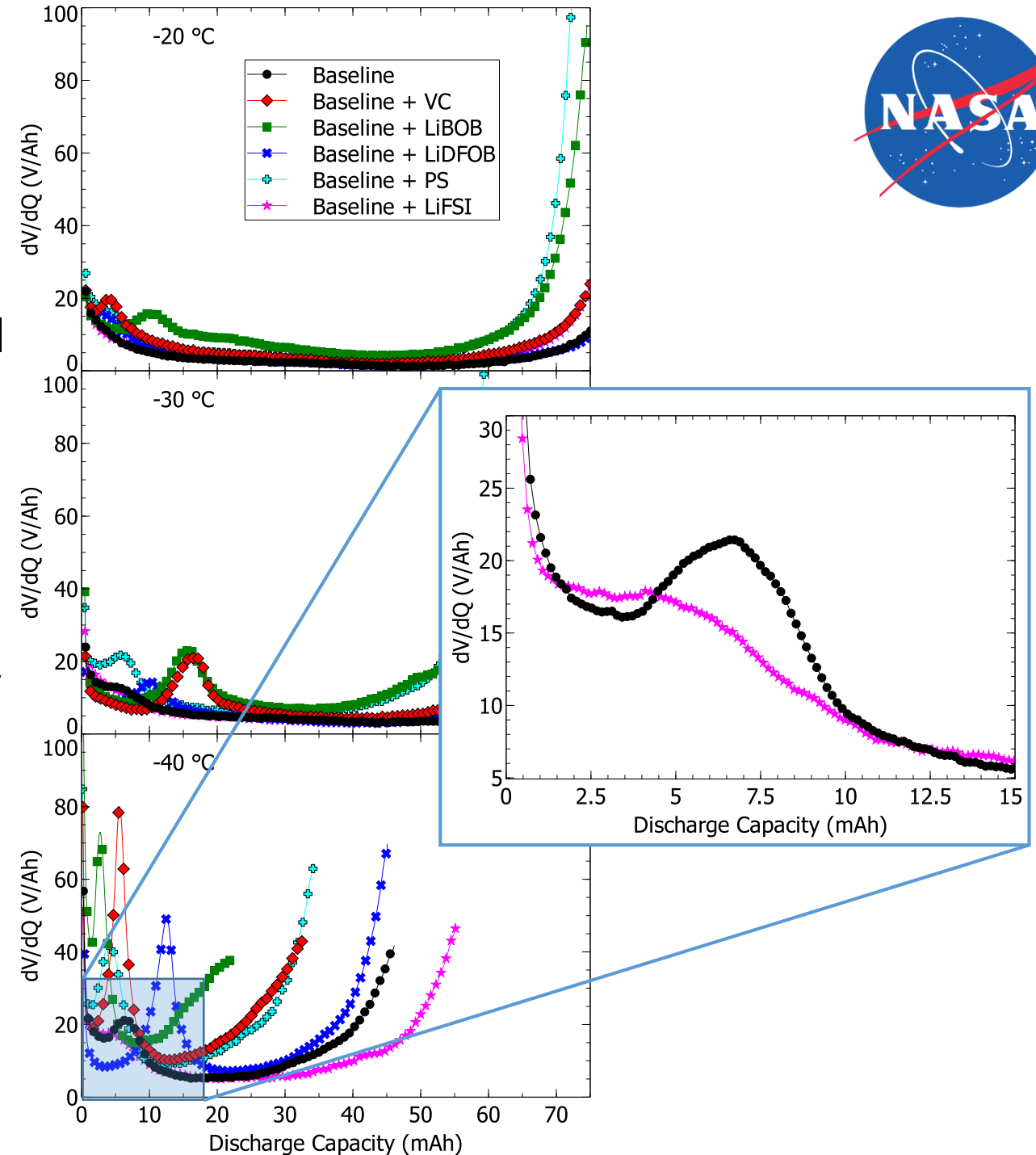
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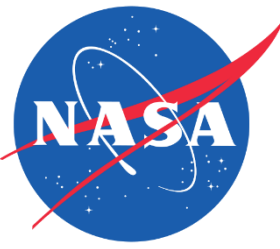




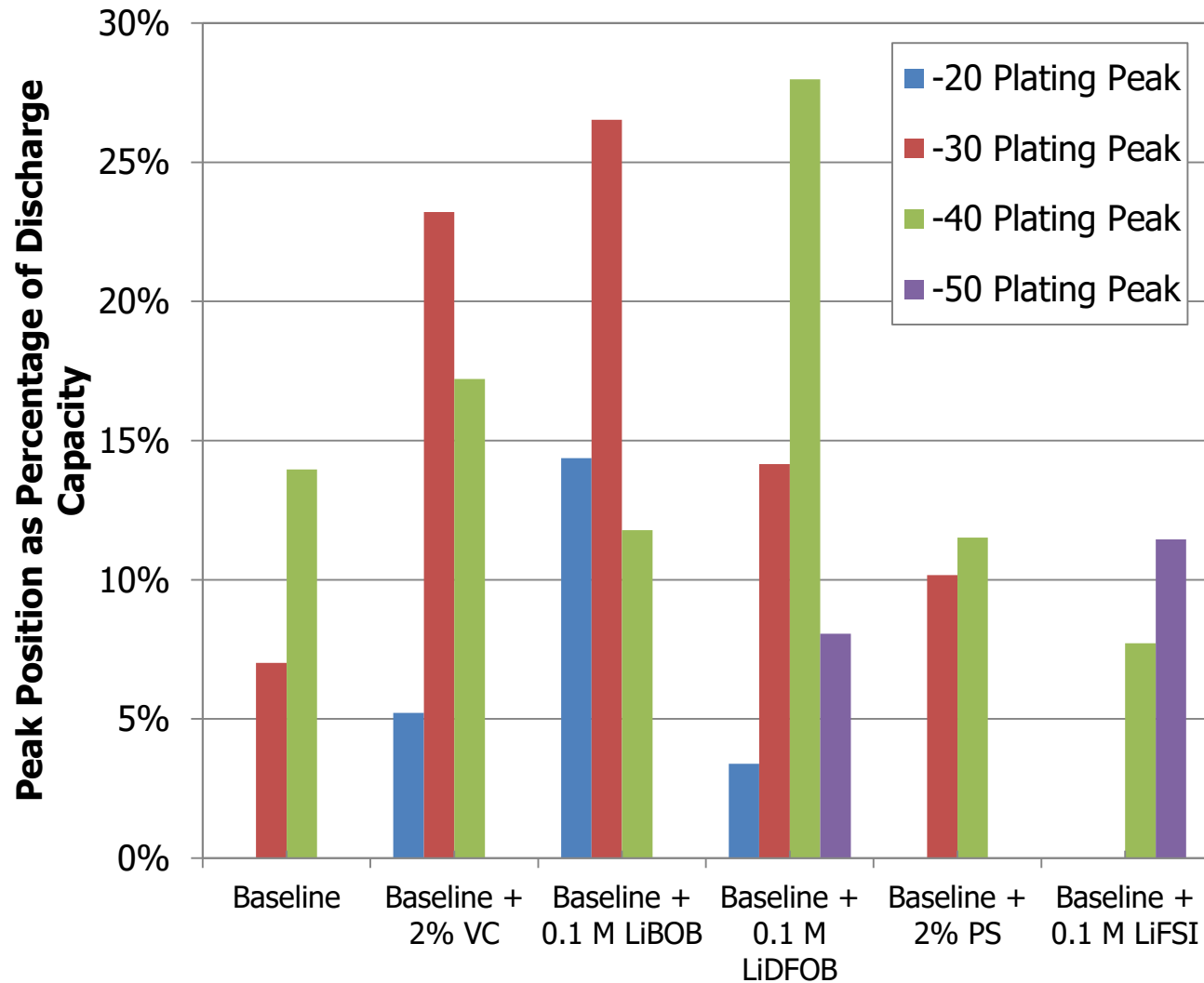
# Comparison of Differential Capacity of Anodes at -20, -30, -40 °C

- No plating observed for baseline until -40 °C
- Cells with VC and LiBOB show evidence of plating at -20 °C and below
  - Lower temperature does not necessarily lead to increased plating
  - Cell with LiBOB produces substantially less plating at -40 °C than -30 °C
- Cells with PS and LiDFOB show plating at -30 °C
- Cell with LiFSI shows minimal evidence for plating even at -40 °C





# Estimating Plating



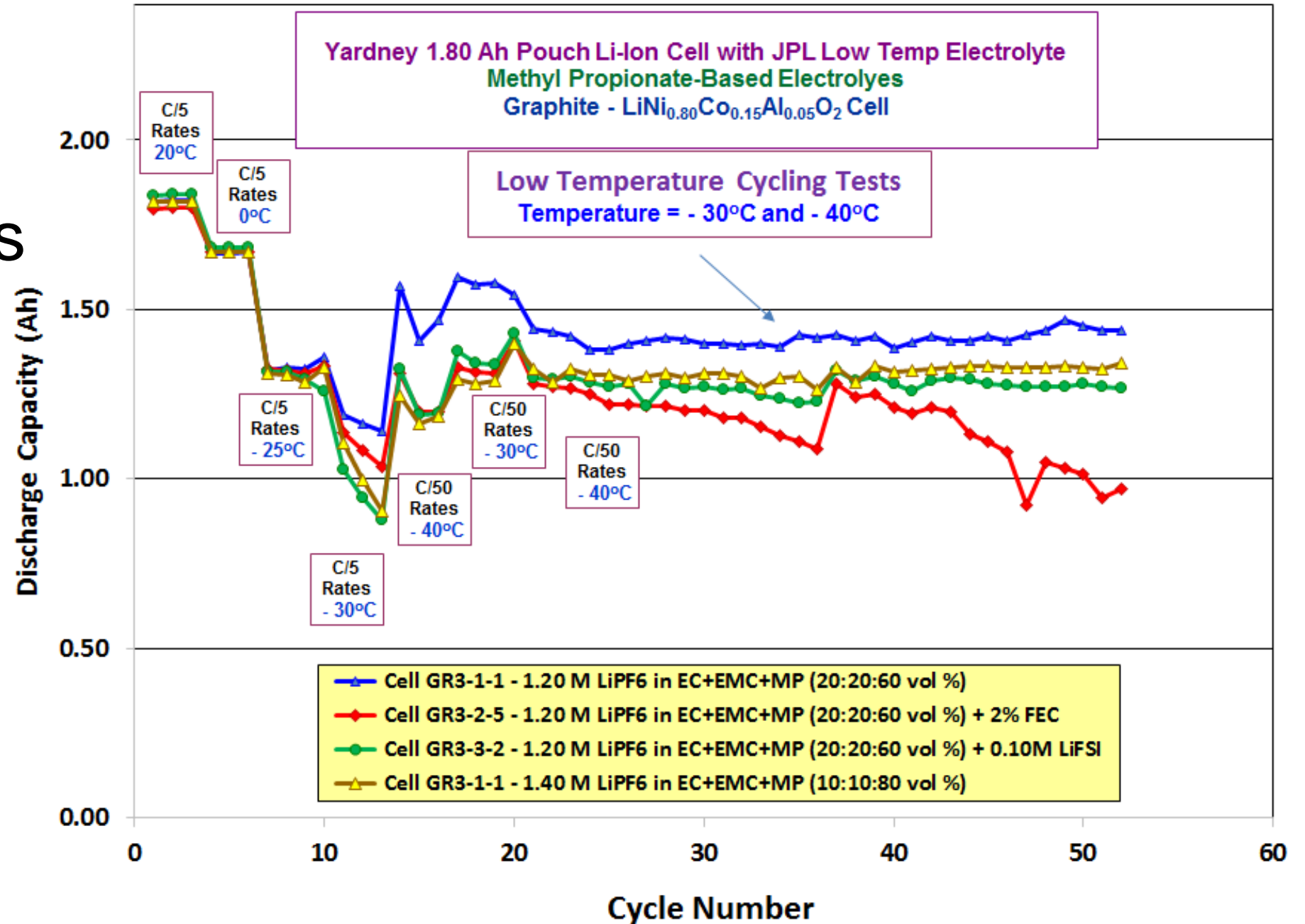
- **Assume** plated lithium is discharged first at low rates (C/20)\*
  - Lithium stripping more facile than de-intercalation
- Peak position at maximum divided by total delivered capacity at that temperature
- Most electrolytes plate less lithium below a certain temperature
- LiFSI electrolyte displays significantly better performance than baseline during C/5 charging at low temperature
- -50 °C C/5 charging possible with only 11 % of capacity stored as plated lithium
- Despite good electrode kinetic data, LiDFOB does not perform better than baseline during plating
  - 28% lithium plating at -40 °C
- Other additives are all far worse than baseline

\*Ref: M. Petzl and M. A. Danzer, *J. Power Sources*, **254**, 80 (2014).



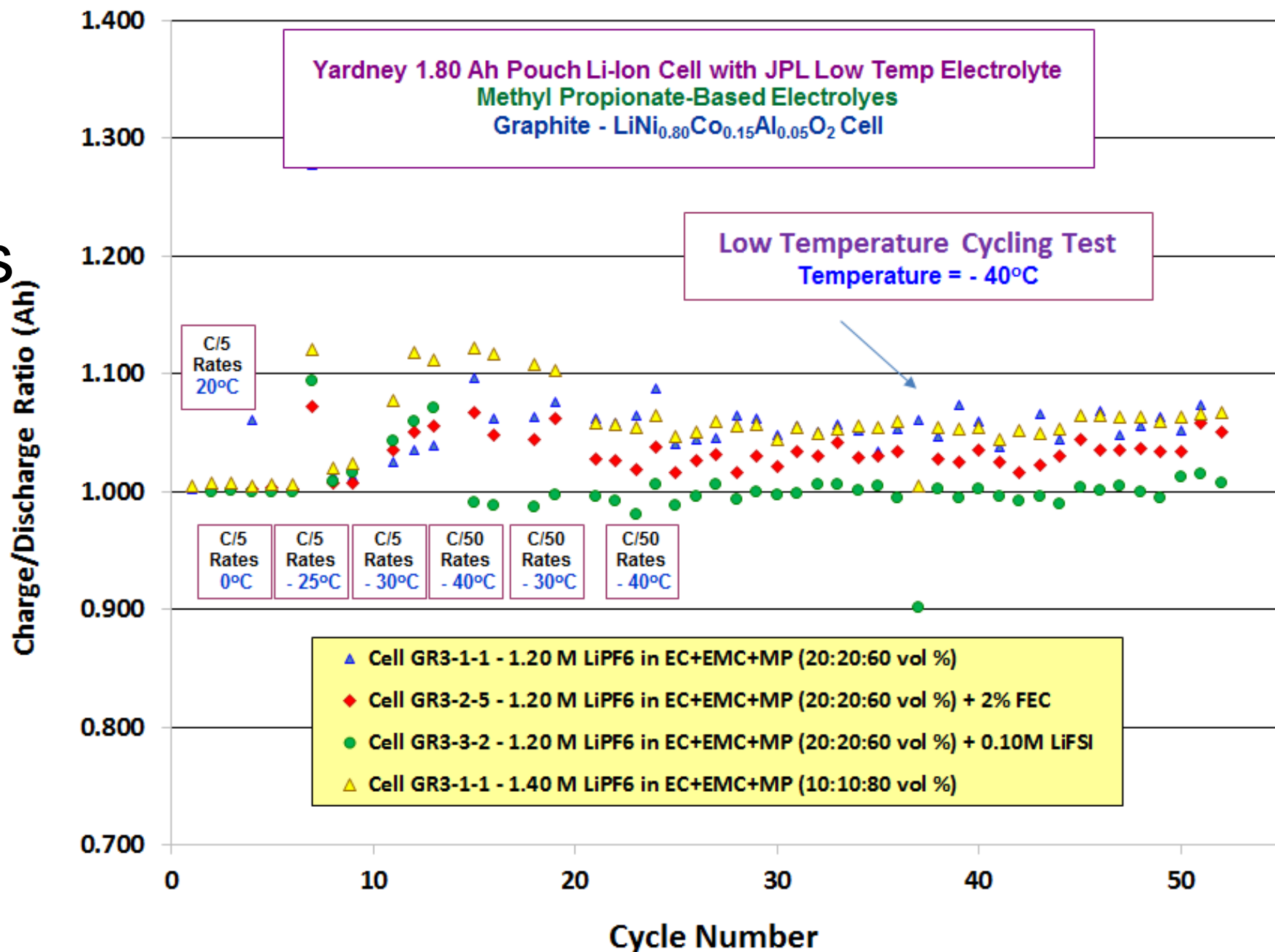
# Evaluating New Electrolytes in Prototype Cells

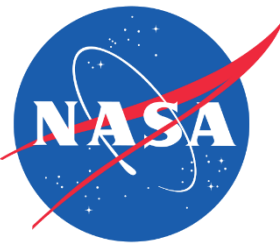
Assessing the Charge Characteristics at Low Temperatures



# Evaluating New Electrolytes in Prototype Cells

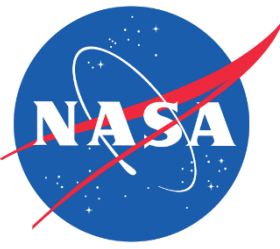
Assessing the Charge Characteristics at Low Temperatures





# Conclusions

- Li/CF<sub>x</sub> batteries can provide **>730 Wh/kg**
  - More than double Li/SOCl<sub>2</sub>
  - Poor low temperature/high rate performance
- Improved Li/CF<sub>x</sub>-MnO<sub>2</sub> hybrid batteries can provide **>550 Wh/kg**
  - Able to discharge at -40 °C
  - Compromise between capacity and low temperature/high rate
  - Appear to sustain gamma radiation without significant energy loss
- JPL-developed Li-ion electrolytes improve low temperature performance
  - Lithium plating can be an issue during charging
  - Improved lithium plating with 0.1 M LiFSI



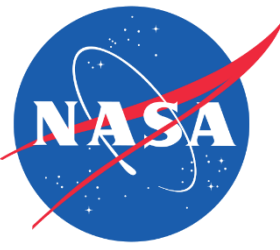
# Acknowledgements

## People

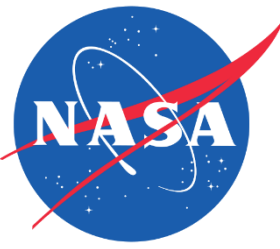
- Marshall Smart
- Simon Jones
- Ratnakumar Bugga
- Erik Brandon
- Charlie Krause
- Will West
- Jasmina Pasalic
- Keith Chin
- Ray Ontiveros
- Keith Billings
- Adam Lawrence
- Candace Seu
- Loraine Torres

## Funding

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# Backup



# Types of Batteries

- Primary (single use)

- Zn/MnO<sub>2</sub> (Alkaline)
- Li/MnO<sub>2</sub>
- Li/SOCl<sub>2</sub>
- Li/SO<sub>2</sub>
- **Li/CF<sub>x</sub>**
- **Li/CF<sub>x</sub>-MnO<sub>2</sub>**

Not reversible

## Examples

Li anode:  
 $\text{Li} \rightarrow \text{Li}^+ + \text{e}^-$

MnO<sub>2</sub> cathode:  
 $\text{MnO}_2 + \text{Li}^+ + \text{e}^- \rightarrow \text{LiMnO}_2$



CF<sub>x</sub>  
Cathode



CF<sub>x</sub>-MnO<sub>2</sub>  
Cathode

- Secondary (rechargeable)

- Lead acid
- Ni/H<sub>2</sub>
- Ni/Cd
- Ni/MH (nickel metal hydride)
- **Li-ion** (various)

Reversible

Carbon anode:  
 $\text{LiC}_6 \leftrightarrow \text{C}_6 + \text{Li}^+ + \text{e}^-$

LiCoO<sub>2</sub> cathode:  
 $\text{Li}_{1-x}\text{CoO}_2 + x\text{Li}^+ + \text{e}^- \leftrightarrow \text{Li}_x\text{CoO}_2$